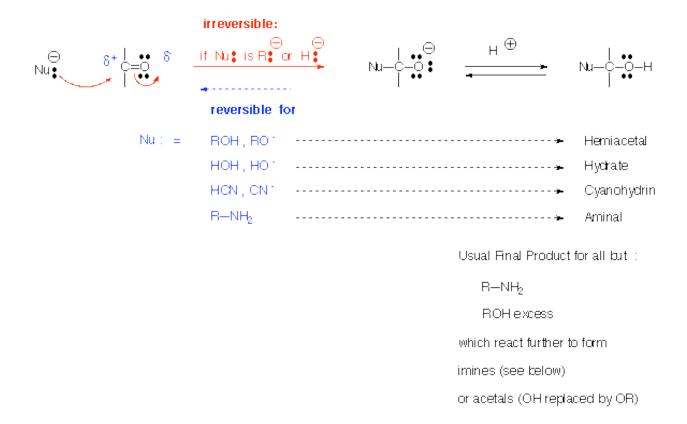
## **REACTIONS OF ALDEHYDES & KETONES**

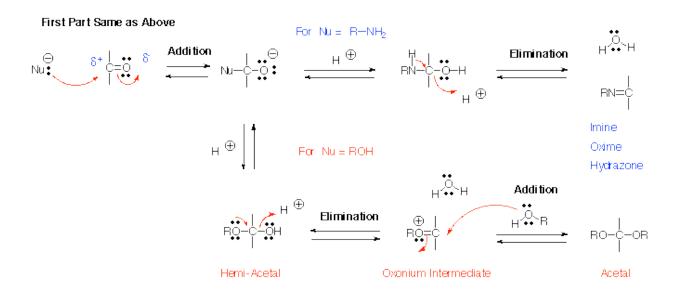
## Most of the reactions of aldehydes and ketones can be classified as:

- 1. Addition to carbonyl by strong (irreversible) or weak (reversible) nucleophiles
- 2. Addition to carbonyl followed by an elimination (RNH2 and Wittig)
- 3. Reaction at alpha carbon with an electrophile

## 1. Addition to Carbonyl



## 2. Addition to Carbonyl Followed by Elimination (for RNH2 and ROH Excess)



### 3. Wittig Reaction (also an Addition-Elimination)

Wittig Reagent can be formed by:

Overall Transformation:

# 4. Reaction at Alpha Carbon

Base: 
$$\begin{array}{c} 8^+ \\ 8^+ \\ \hline \\ 8^+ \\ \hline \end{array}$$
 Base—H  $\begin{array}{c} 0^+ \\ 0^+ \\ \hline \\ \end{array}$  Base—H  $\begin{array}{c} 0^+ \\ 0^+ \\ \hline \\ \end{array}$  Base—H  $\begin{array}{c} 0^+ \\ 0^+ \\ \hline \\ \end{array}$  Base—H  $\begin{array}{c} 0^+ \\ 0^+ \\ \hline \\ \end{array}$  Base—H  $\begin{array}{c} 0^+ \\ 0^+ \\ \hline \\ \end{array}$  Base—H  $\begin{array}{c} 0^+ \\ 0^+ \\ \hline \\ \end{array}$  Base—H  $\begin{array}{c} 0^+ \\ 0^+ \\ \hline \end{array}$  Base—H  $\begin{array}{c} 0$