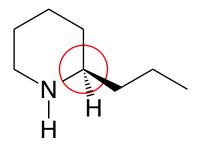
CHEM 261 Oct 21, 2015

### **Previous lecture: Enantiomers of coniine**



## **Determining R/S configuration of the stereocenter:**

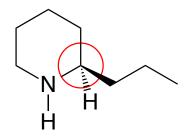
$$\begin{array}{c}
C \\
C \\
1 \\
N \\
H \\
4
\end{array}$$

$$\begin{array}{c}
C \\
C \\
H \\
H \\
H \\
H \\
H
\end{array}$$

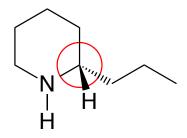
$$\begin{array}{c}
C \\
C \\
H \\
H \\
H \\
H
\end{array}$$

- We can assign highest priority to the N and lowest to the H, but cannot immediately tell which carbon attached to the stereocenter is of higher/lower priority. When this is the case, we look at the next substituents in the chain.

- We cannot tell at the second attached carbon, so we move on to the third.
- We still cannot tell at the third, so we move on to the fourth.
- At the fourth carbon we can see a difference. The carbon that is part of the propyl group ends in a CH<sub>3</sub> so it is bonded to three H, and the other carbon is bonded to two H and one C. The propyl group gets lower priority (3) and the other group gets higher priority (2).
- Counting 1,2,3  $\rightarrow$  clockwise is R. This is the R enantiomer.



R - enantiomer of coniine



S - enantiomer of coniine - invert EVERY stereocenter

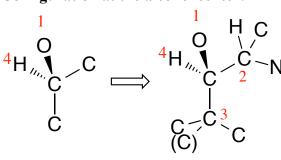
#### **Chiral Centers:**

quinine - anti-malarial drug from the bark of the tree *Cinchona officinalis* 

malaria is cause by *Plasmodium* species transmitted by *Anopheles* mosquito

- Carbon stereocenters are shown with dots in this example.
- Nitrogen is a stereocenter here because it can't invert freely. The ring structure restricts its geometry.

#### Configuration at the alcohol center:



- Cannot assign 2, 3 at first try
- At the second atoms in the chain, there is a difference. The carbon attached to one nitrogen, one carbon, and one hydrogen has a higher priority than the carbon attached to three carbons.
- One nitrogen trumps three carbons.
- Count 1, 2, 3: Clockwise is R.

- Morphine (from Morpheus, Greek god of sleep)
- 5 stereogenic centers
- Opium: Sap from the seed pod of opium poppy (*Papaver somniferum*)
- $\sim$ 10% of opium is morphine
- Morphine is used as an analgesic
- Heroin (diacetylmorphine) is even more potent (and more addictive)

#### Configuration at the ether stereocenter:

$$\begin{array}{c}
1 & C \\
0 & C \\
4H & C \\
C & C
\end{array}$$

$$\begin{array}{c}
1 & C \\
0 & C \\
C & 3 \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

$$\begin{array}{c}
C & C \\
C & C
\end{array}$$

- Cannot assign 2, 3 at first try
- At the second atoms in the chain, there is a difference. The alcohol carbon is attached to one oxygen, one carbon, and one hydrogen. It has a higher priority than the other carbon which is attached to three carbons.
- Count 1, 2, 3: Counterclockwise
- This center is *R* and not *S* because the lowest priority group (the hydrogen) is pointing toward the front, not to the back.

# Configuration at the alcohol stereocenter:

- Cannot assign 2, 3 at first try
- At the second atoms in the chain, the ether carbon is attached to one oxygen, one carbon, and one hydrogen. It has a higher priority than the alkene carbon which is attached to two carbons and one hydrogen.
- Count 1, 2, 3: Clockwise
- This center is *S* and not *R* because the lowest priority group (the hydrogen) is pointing toward the front, not to the back.