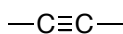


ALKENES AND ALKYNES

Alkenes and Alkynes – Term olefin comes from: oleum facere

▶ Olefin

“oil” + “to make”



Alkene (olefin) Alkyne (acetylene)

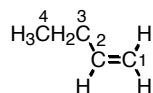
Alkenes – structure and nomenclature



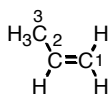
ethylene (common name)

ethene (systematic name)

replace "ane" of corresponding
alkane name with "ene"



1-butene

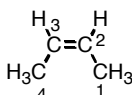
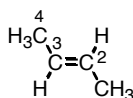


propylene (common name)

or propene(systematic name)

- to name find longest chain containing maximum number of C=C with both multiply bonded carbons in chain

- number from end to give 1st carbon of C=C lowest number,
prefix with number to indicate position of first double bonded carbon



- cis = two large groups on same side

- trans = two large groups on opposite side

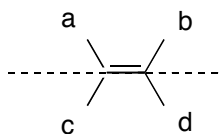


1
trans-2-butene

cis-2-butene

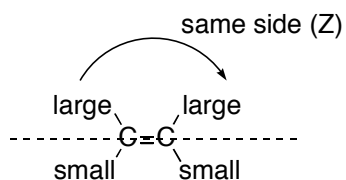
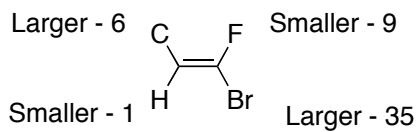
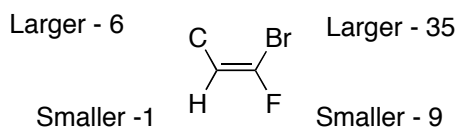
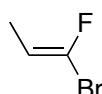
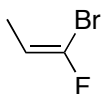
These 2-butenes are structural isomers with respect to the 1-butene above

Stereoisomers (Diastereomers)

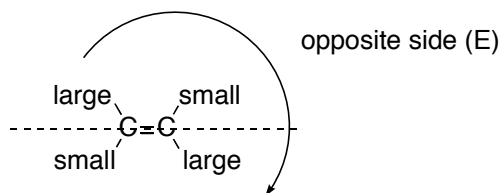


- examine atomic no.'s of elements directly attached to double bond
- compare a and c
- compare b and d
- determine if largest groups are on same side or opposite sides

Example 1

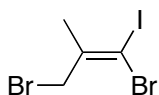


Z-1-bromo-1-fluoro-1-propene

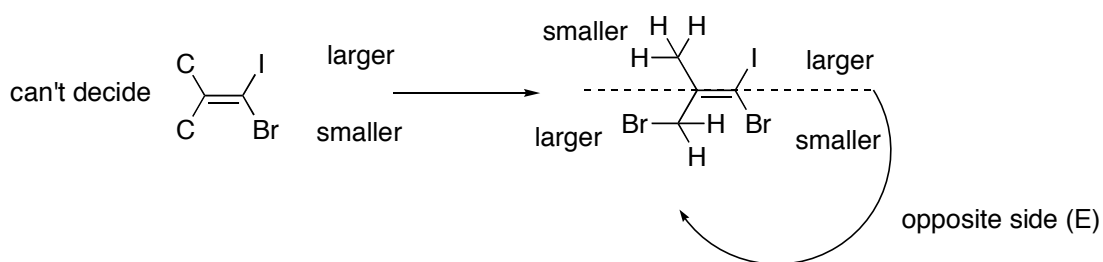


E-1-bromo-1-fluoro-1-propene

Example 2



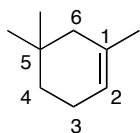
E-1,3-dibromo-1-iodo-2-methyl-1-propene



Nomenclature Alkenes and Alkynes -continued



cyclopropene - ringstrain makes this molecule very reactive

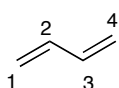


1,5,5-trimethylcyclohexene or
1,5,5-trimethyl-1-cyclohexene

For molecules having multiple double bonds:

- 2 C=C bonds → diene
- 3 C=C bonds → triene
- 4 C=C bonds → tetraene
- 5 C=C bonds → pentaene

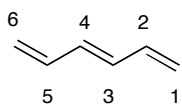
Eg.1



1,3-butadiene

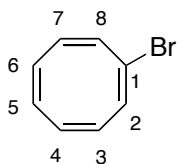
drop "ne" ending of alkane and add "diene"

Eg.2

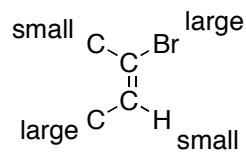


1,3,5-hexatriene

Eg.3

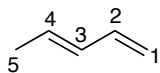


1-E-1-bromo-1,3,5,7-cyclooctatetraene

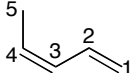


Eg.4

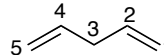
E-1,3-pentadiene



Z-1,3-pentadiene



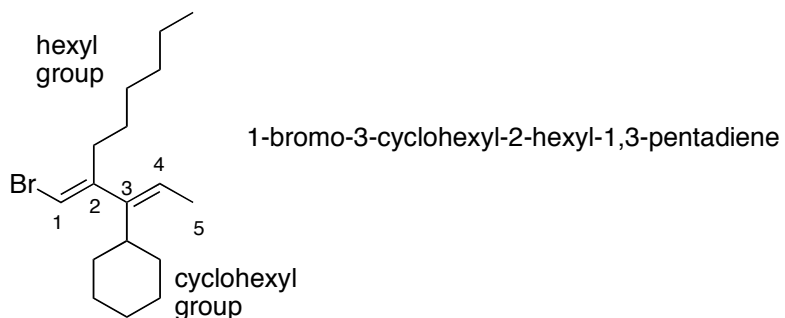
1,4-pentadiene



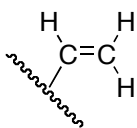
Stereoisomers
(diastereomers)

Structural
isomers

Eg.5



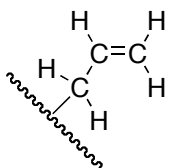
Different groups



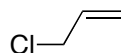
Vinyl group



Vinyl chloride
1-chloroethene



Allyl group



Allyl chloride
3-chloropropene

Natural Alkenes eg. Terpenes/Terpenoids or Isoprenes/Isoprenoids

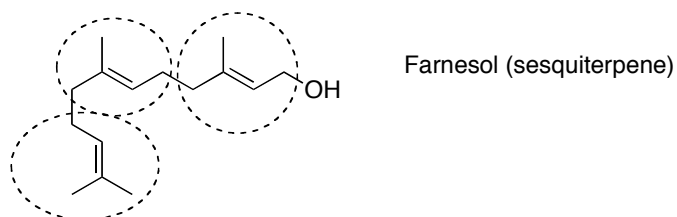
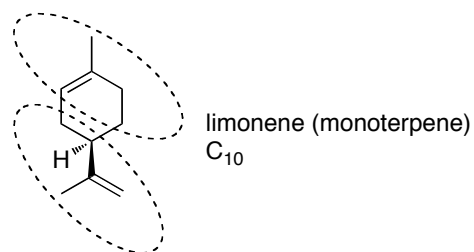
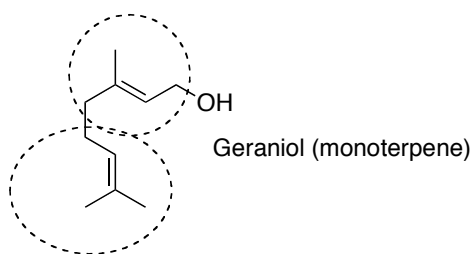


Isoprene
2-methyl-1,3-butadiene



C₅

- 2 C₅ (isoprene units) – monoterpene
- 3 C₅ (isoprene units) – sesquiterpene
- 4 C₅ (isoprene units) – diterpene
- 6 C₅ (isoprene units) – triterpene



Note - Each dotted circle identifies a C₅ unit

Pheromones (chemical messengers)

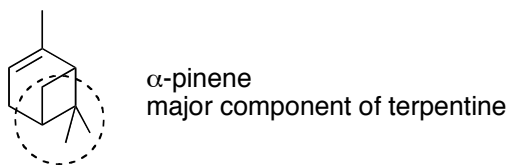
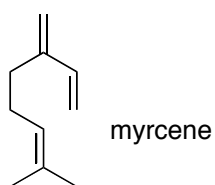
From the Greek	Pherein	Horman
	to carry	excitement

1959 – Adolf Butenandt (Nobel prize for discovery of pheromones)

Sex Pheromones – Insects can detect 10^{-17} moles/L (i.e. 10^{-17} M or 10^{-17} molar)

Alarm Pheromones

Trail Pheromones



Swarm Pheromones

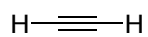
Both of these are monoterpenes

Nomenclature of alkynes

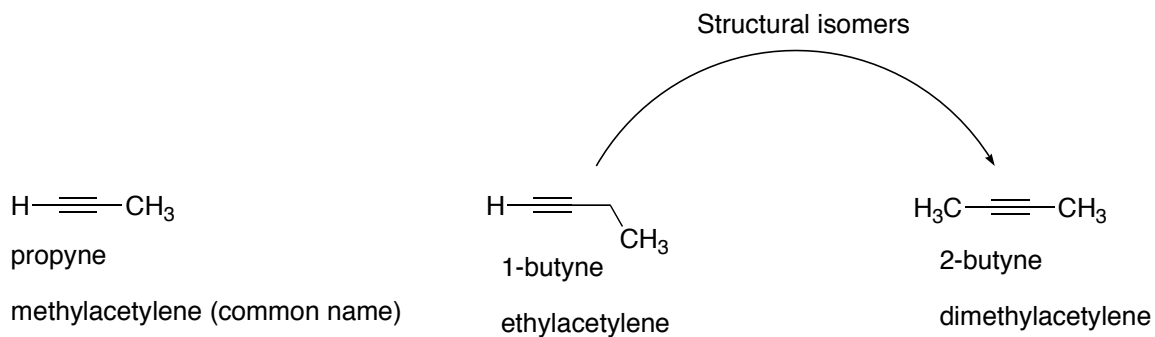
Rule:

- drop “ane” and add “yne”
- for multiple triple bonds, drop “ne” and add “diyne”, “triyne”, etc.

eg.



- ethyne / acetylene (common name)



Multiple alkynes end with -

2 $\text{C} \equiv \text{C}$ diyne

3 $\text{C} \equiv \text{C}$ triyne

Mixed double and triple bond containing compounds are “eneynes”

The below example is from Canola – defense substance – anti-nematode

