

Chem 261
Assignment & Lecture Outline 3:
Stereochemistry, Alkyl Halide Substitution (S_N1 & S_N2)

Read

Organic Chemistry, Solomons, Fryle & Snyder 13th Edition (Electronic)

- Functional Group List – Learn to recognize – Please see Green Handout – also p 75 of text
- Periodic Table – Inside Back Cover - know 1st 10 elements (up through Neon)
- Relative Strength of Acids and Bases – Inside Back cover (reference only)
- Chapter 5 – Stereochemistry
- Chapter 6 –Ionic Reactions: Nucleophilic Substitution of Alkyl Halides

Problems:

Do **Not** turn in, answers available in "Study Guide Student Solutions Manual " Solomons, Fryle, Snyder

- **Chapter 5:** 5.1 to 5.15; 5.18 to 5.21; 5.26; 5.28; 5.33a-d; 5.46
- **Chapter 6:** 6.1 to 6.5; 6.7 to 6.10; 6.13; 6.20; 6.26; 6.27

Lecture Outline # 3

I. Comparison of 2 Structures:

Same Molecular Formula ? -> If Yes, Possibly Isomers or Identical
Same Arrangement (Sequence) of Groups ?

If No -> Structural Isomers

If Yes -> Superposable?

If Yes -> Identical Structures

If No -> Stereoisomers

Non-Superposable Mirror Images ?

If NO -> Diastereomers

If Yes -> Enantiomers

II. Chirality and Stereoisomers

A. The Concept of Chirality

1. Identification of chiral objects
 - a) achiral = not chiral
 - b) planes of symmetry within a molecule
2. Types of stereoisomers – enantiomers and diastereomers

B. Location of stereogenic (chiral) centres – 4 different groups on tetrahedral atom

1. Enantiomers & diastereomers
2. Meso compounds - chiral centers with plane of symmetry within molecule
3. Molecules with more than one chiral centre
4. Recognition of chiral centers in complex molecules - cholesterol - 8 chiral centres
Drawing the enantiomer of cholesterol and its potential 255 stereoisomers
5. Fischer projections

C. R and S nomenclature

1. Rules for Assigning R and S configurations
2. Treatment of multiple bonds

III. Optical Rotation, Optical Purity, and Resolution of Enantiomers

A. Optical Rotation

1. Measurement, factors, and absolute rotation
2. Optical purity and enantiomeric excess
3. Physical Properties of Enantiomers and Diastereomers
4. Racemic mixtures - 50-50 mixtures of enantiomers
5. Optical Purity = enantiomeric excess (ee)

B. Separation (Resolution) of Enantiomers (e.g. Racemic mixtures)

1. Creation of diastereomers
2. Biological recognition

IV. Nucleophilic Substitution Reactions (S_N1 and S_N2)

A. General Features of Nucleophilic Substitution vs. Elimination Reactions

1. Definitions: S_N1 and S_N2
2. Mechanisms

B. S_N2 Reactions

1. Stereochemistry - Walden Inversion (inversion of configuration)
2. Substitution of Primary and Secondary Alkyl Halides
3. Synthesis of alcohols, ethers, other halides, etc.

C. S_N1 Reactions

1. Stereochemical Aspects (loss of stereochemistry via carbocations)
2. Substitution of Tertiary Alkyl Halides and Other Tertiary Carbons
3. Synthesis of alcohols, ethers, halides