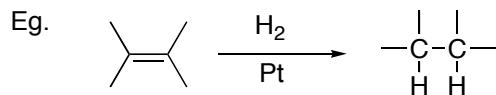


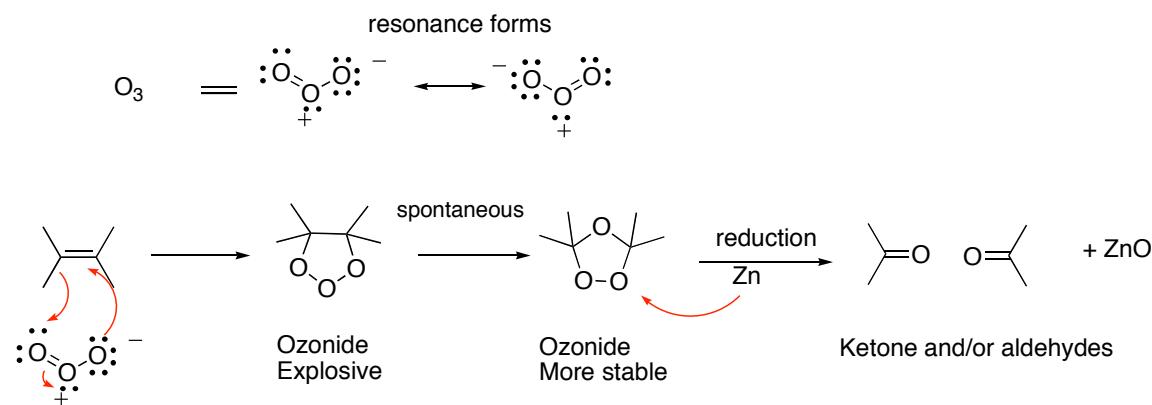
Reduction: process that adds electrons



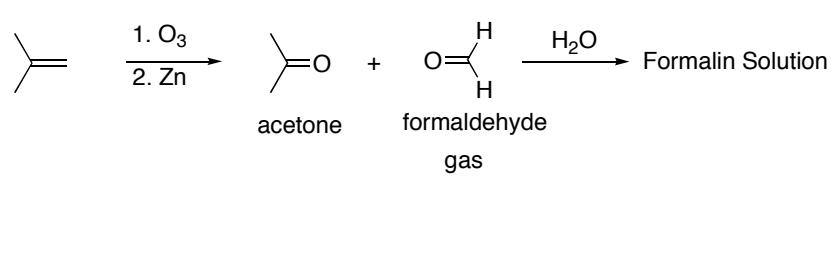
Oxidation: process that removes electrons

Ozonolysis: cleavage of alkenes by ozone (O_3)

General reaction:

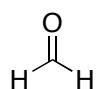


Eg.

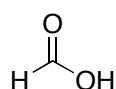


$C=O$	carbonyl
$\begin{array}{c} C \\ \\ C=O \end{array}$	ketone
$\begin{array}{c} (H)C \\ \\ C=O \\ \\ H \end{array}$	aldehyde

formaldehyde name comes from formic acid, which comes from formica (ant):

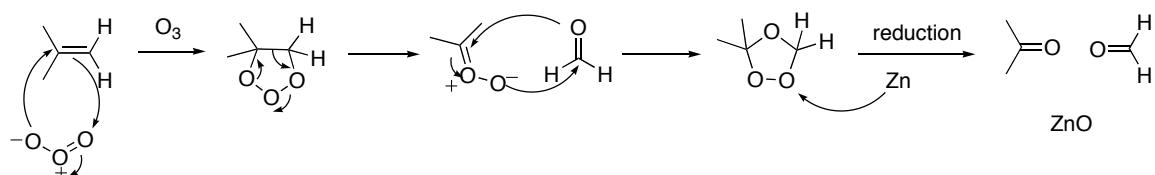


formaldehyde



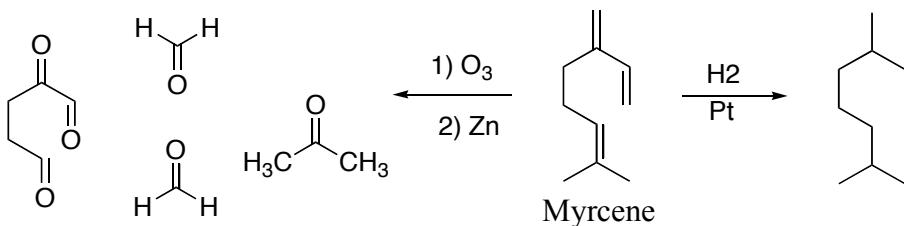
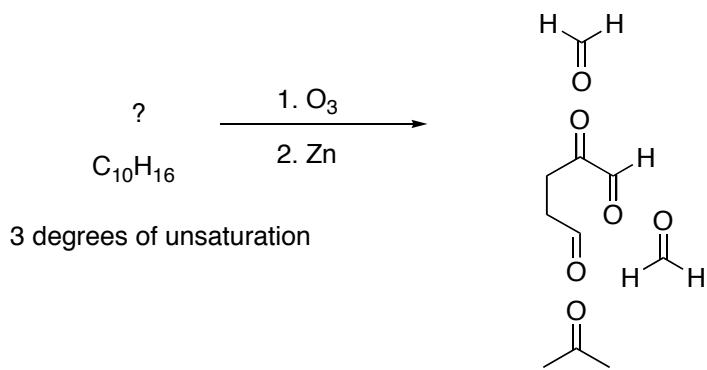
formic acid is an ant defense substance
(formica means ant)

Mechanism:



Sample question for practice: What is a possible structure for the starting material below ($\text{C}_{10}\text{H}_{16}$)?

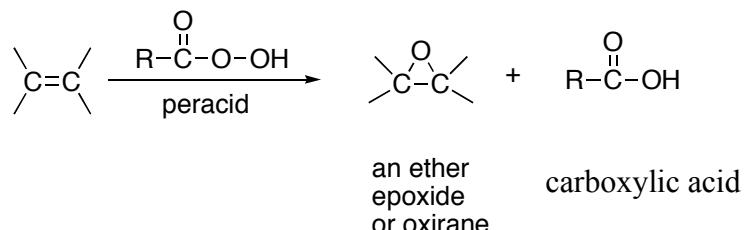
Are there other isomers that will give the same products for ozonolysis followed by Zn treatment?



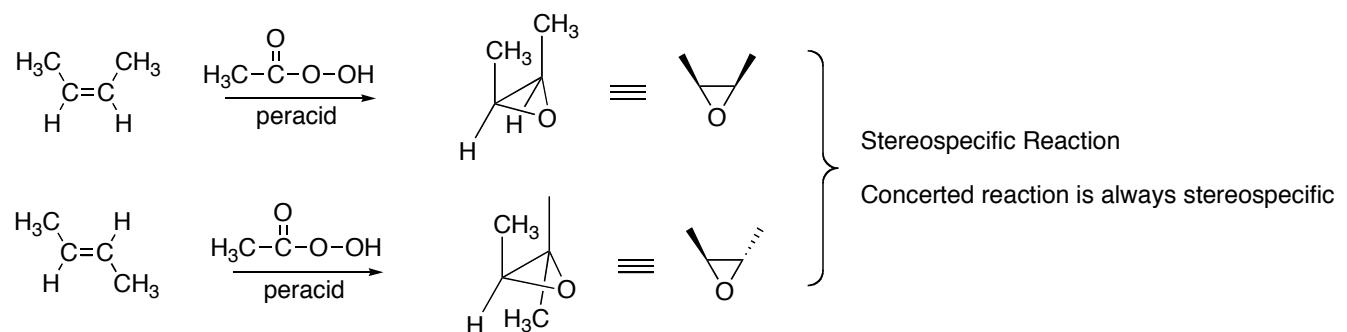
Epoxidation – Oxirane formation

1) Epoxide formation (epoxidation) – Oxirane formation

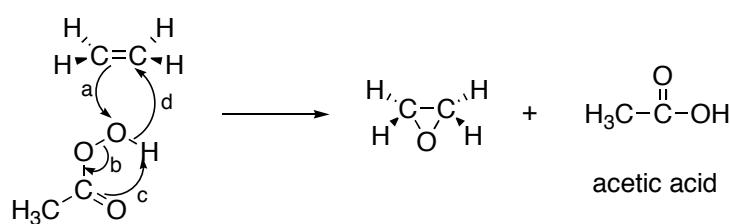
ex)



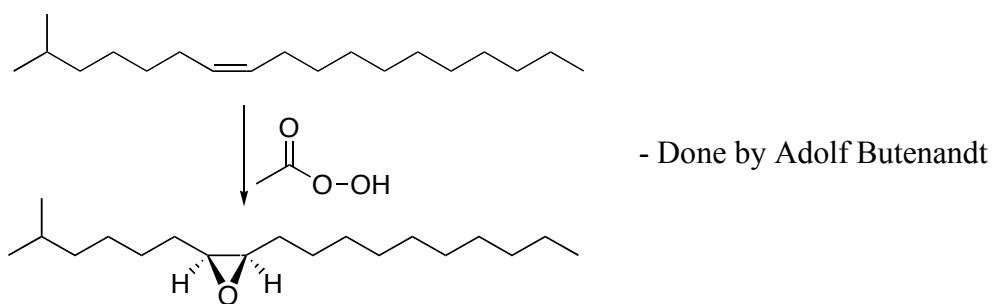
- Syn Addition
- Concerted reaction: all bonds break and form at the same time



Mechanism:



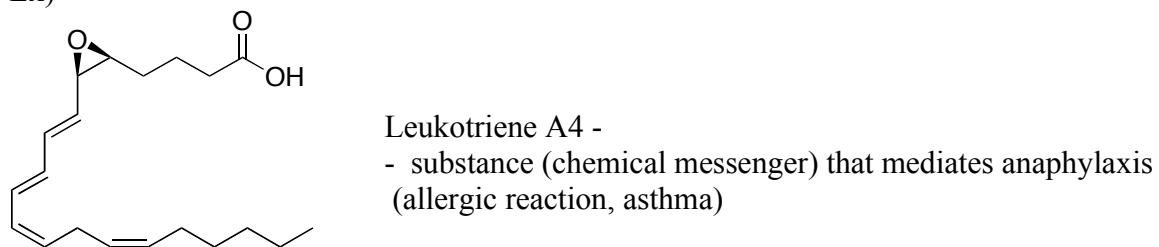
ex)



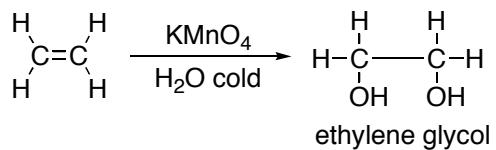
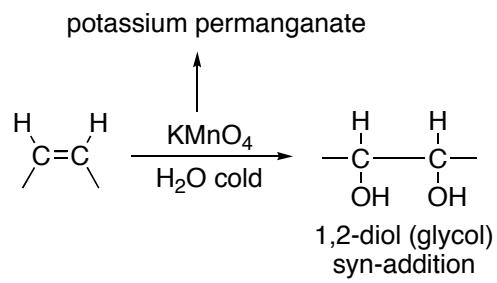
Gypsy moth sex pheromone

- Pheromones (pheromone – to carry excitement)
- Chemical messengers

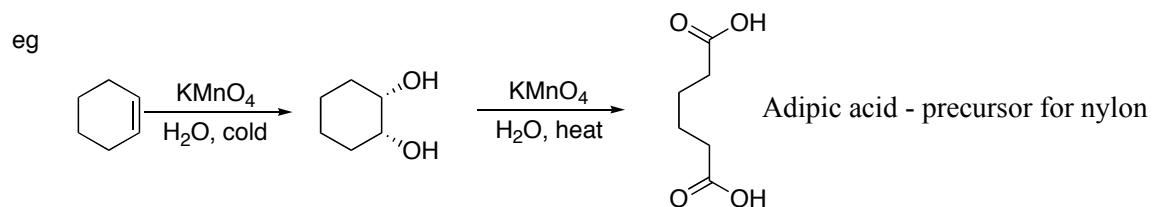
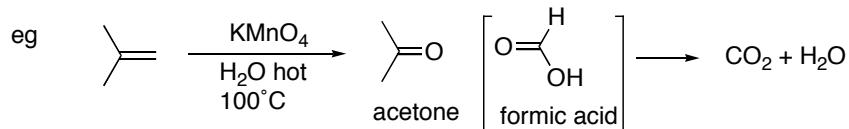
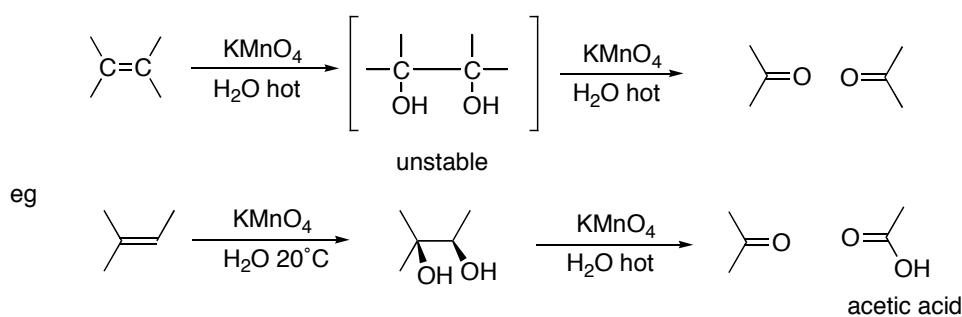
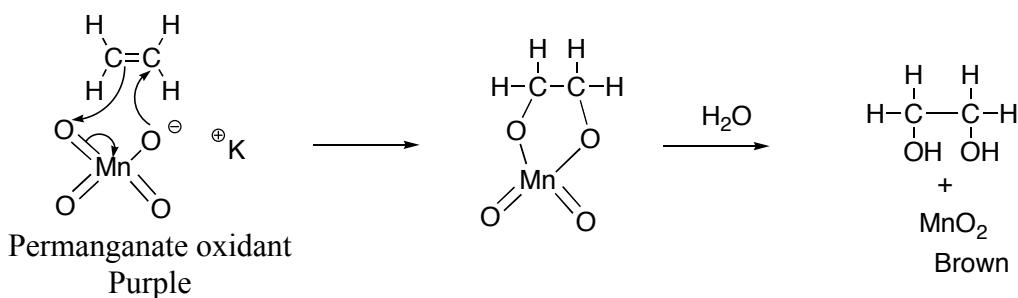
Ex)



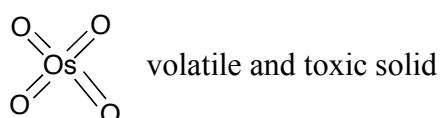
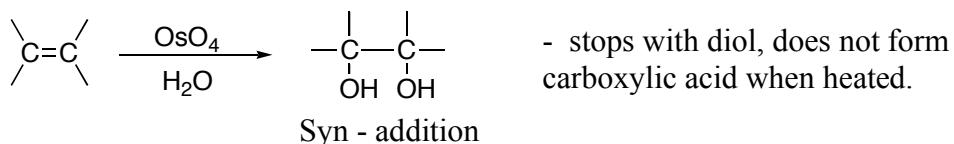
Dihydroxylation



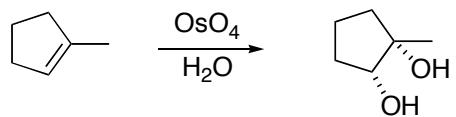
Mechanism



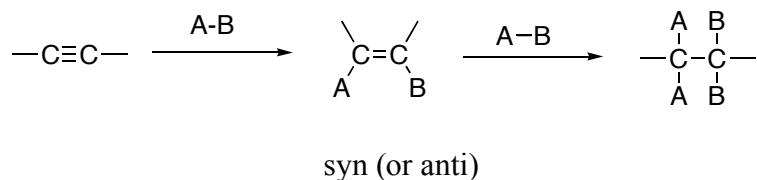
OsO₄, osmium tetroxide.



eg.

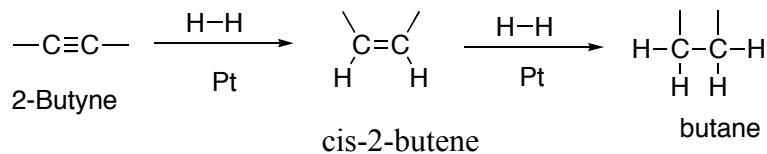


Addition Reactions of Alkynes



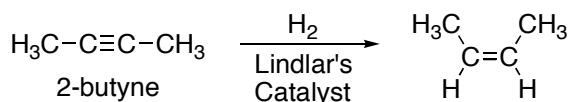
1) Hydrogen Addition

Ex.

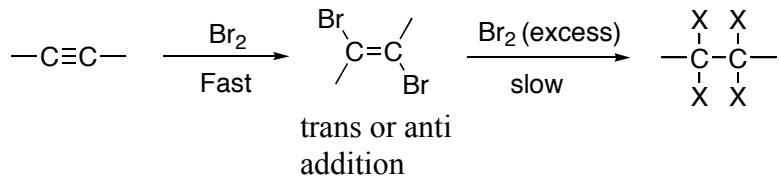


reaction can not be easily stopped at cis alkene

- to stop at alkene use Lindlar's catalyst: Pd, BaSO₄ or CaCO₃ quinoline



2) Halogenation:



Two Br are pulling electrons away,
and slow down the second step