GLUCOSE HEMI-ACETAL & ACETAL (GLYOSIDE) FORMATION: SOME COMMON CONCEPTS IN CARBOHYDRATE ("SUGAR") CHEMISTRY

Carbohydrates

• (*carbon* + *hydrate*) are molecules with three or more carbons and a general formula that approximates $C_nH_{2n}O_n$

Saccharides

- are carbohydrates or sugars
- monosaccharides have one sugar moiety
- **disaccharides** have two sugars linked together
- **trisaccharides** have three sugars linked together
- tetrasaccharides have four sugars linked together, etc...
- **polysaccharides** have an indeterminate number which can be hundreds of thousands or more

Prefixes and Suffixes

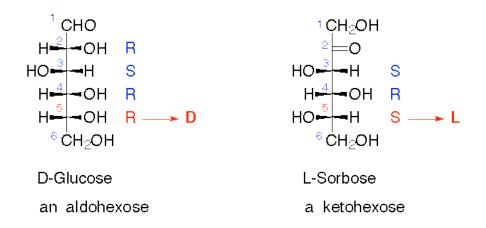
- the suffix (ending) for sugar names is: -ose
- the prefix defines the number of carbons:
 - triose (3 carbons)
 - **tetr**ose (4 carbons)
 - **pent**ose (5 carbons)
 - **hex**ose (6 carbons), etc...
- a further prefix defines the types of carbonyl group in the sugar:
 - o **aldo-** (aldehyde) or
 - **keto-** (ketone)
 - for example glucose (shown below) is an "aldohexose" whereas fructose is a "ketohexose"
- the term **pyranose** means a six-membered sugar ring (hemiacetal or acetal see below)
- the term **furanose** means a five-membered ring
- These terms are often prefixed as in "glucopyranose" which means glucose cyclized to its six-membered ring form (see below)

D- and L- Sugars

- This is a naming convention
- Using standard nomenclature numbering, determine the configuration (R or S) of the

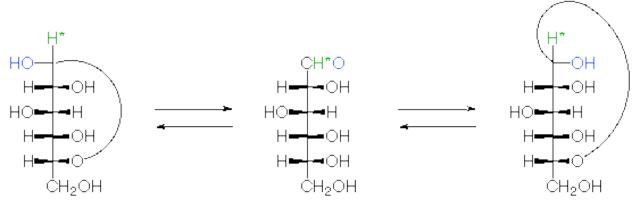
highest numbered stereogenic center ("chiral center" or "asymmetric center"):

- $\circ~$ if it has R-configuration, the sugar is a D-sugar
- o if it has S-configuration, the sugar is an L-sugar



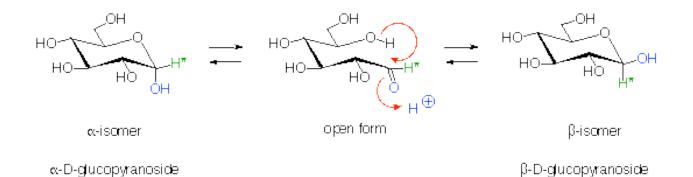
Glucose Hemi-Acetal Formation

- The open form of D-glucose (and many other sugars) can cyclize to form hemiacetals.
- Under acidic conditions the hemiacetal form of glucose can react with other alcohols to give acetals known as **glycosides**. These are widely distributed in nature.
- These open form and cyclized structures can be depicted in different ways. All of these structure can be referred to as "D-glucose"

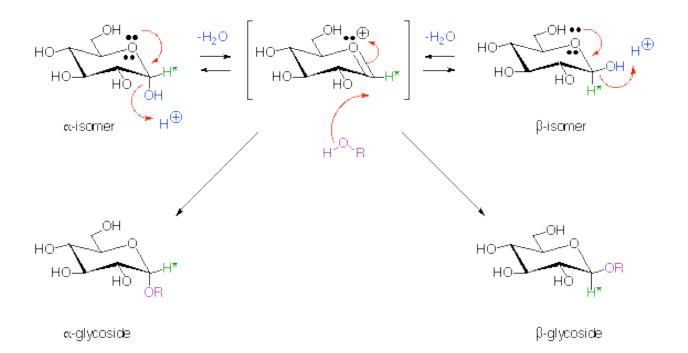


• Fischer Projection:

• **3D Projection:**

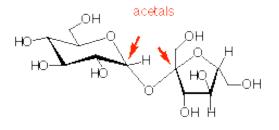


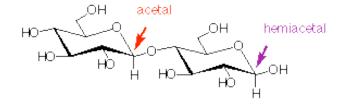
Glucose Acetal (Glycoside) Formation



Examples of Disaccharides

- In the following structures, the anomeric carbons (acetal or hemiacetal) are indicated by coloured arrows.
- The full systematic names of the sugars are given below their common names.





Sucrose (Table Sugar)

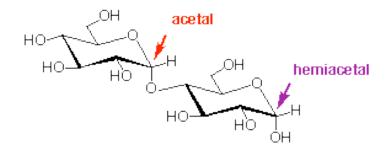
Cellobiose (from Cellulose)

 α -D-glucopyranosyl- β -D-fructofuranoside

a non-reducing sugar

 $\beta\text{-}D\text{-}glucopyranosyl-}\beta\text{-}D\text{-}glucopyranoside}$

a reducing sugar



Maltose (a disaccharide made from starch)

α-D-glucopyranosyl-α-D-glucopyranoside

a reducing sugar