

Redox-gated Molecular Memory Devices based on Dynamic Doping of Polythiophene

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Canada Foundation for Innovation

Today's solid state memory (>\$100 billion annually):

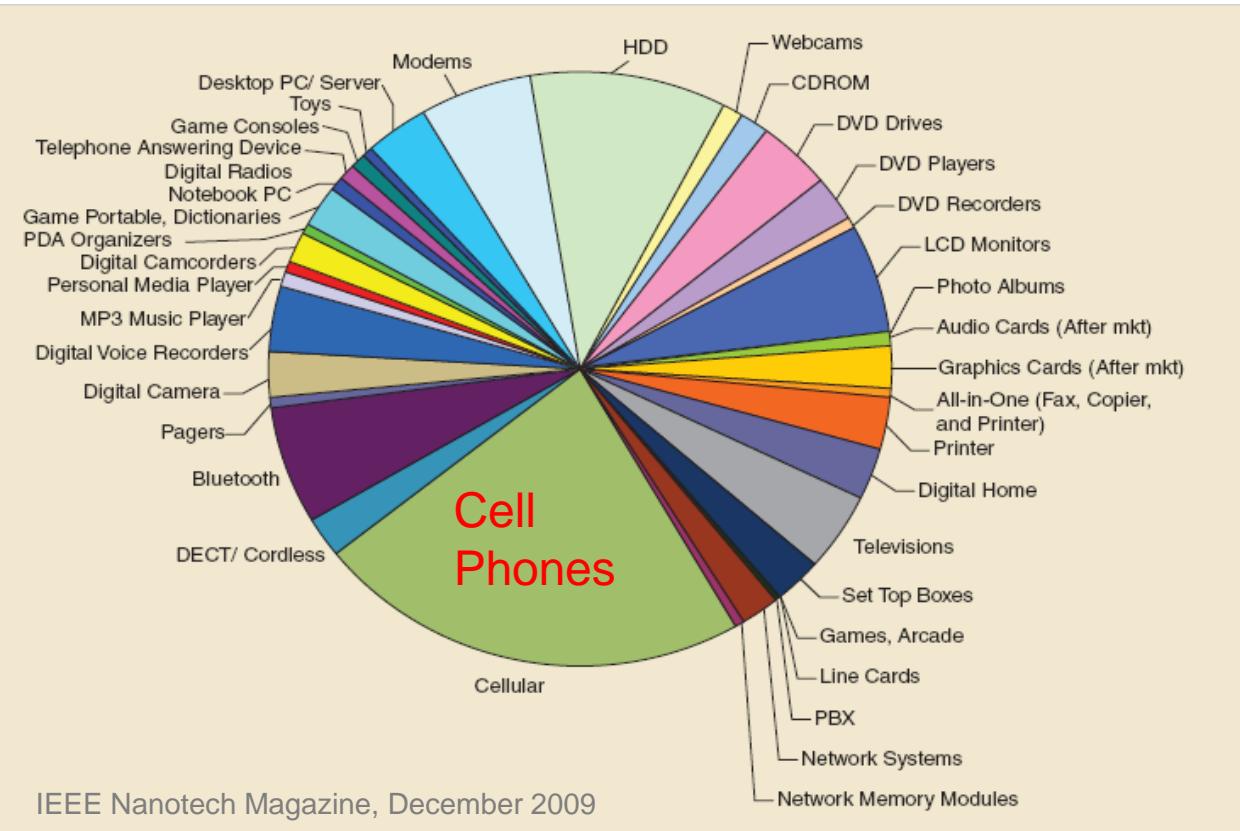
	write/erase speed	cell size	retention	cycle life
Dynamic Random Access Memory (DRAM)	< 10 nsec	1T1C*	65 msec	> 10^{15}
Static RAM	< 1 nsec	6T	long with power on	> 10^{15}
Flash (cameras, USB stick)	1 μ sec- 10 msec	1T	> 10 yrs	$10^3 - 10^5$

No “universal” memory- we need different types for different needs

* T = transistor, C = capacitor

Nonvolatile solid state Memory Devices:

- enabled mobile electronics (cell phones, MP3 players, etc)
- ~\$65 billion/year (not counting disk drives)
- 16 billion units (i.e. chips) sold in 2009



“Alternative” nonvolatile memory in ITRS roadmap:

Phase Change RAM

Magnetic RAM

Polymer memory

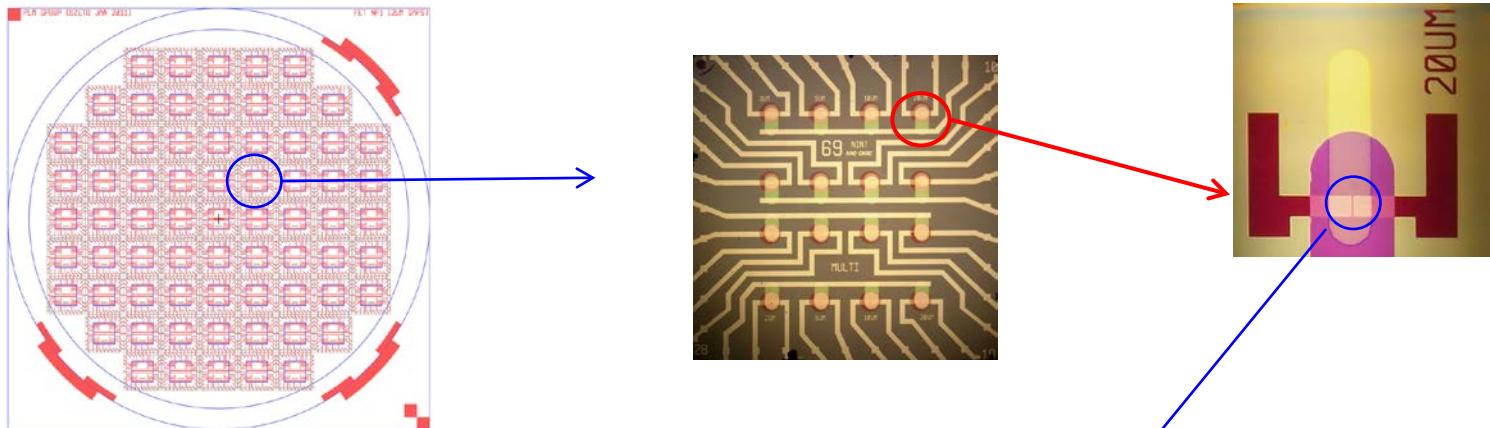
Fuse/antifuse

Molecular memory

Nanomechanical

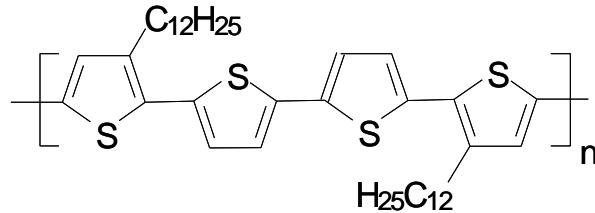
Scifinder hits:
“nonvolatile memory”: 31,800
“resistance memory”: 9,900
“conductance switching”: 10,560

“Redox gated” molecular memory:

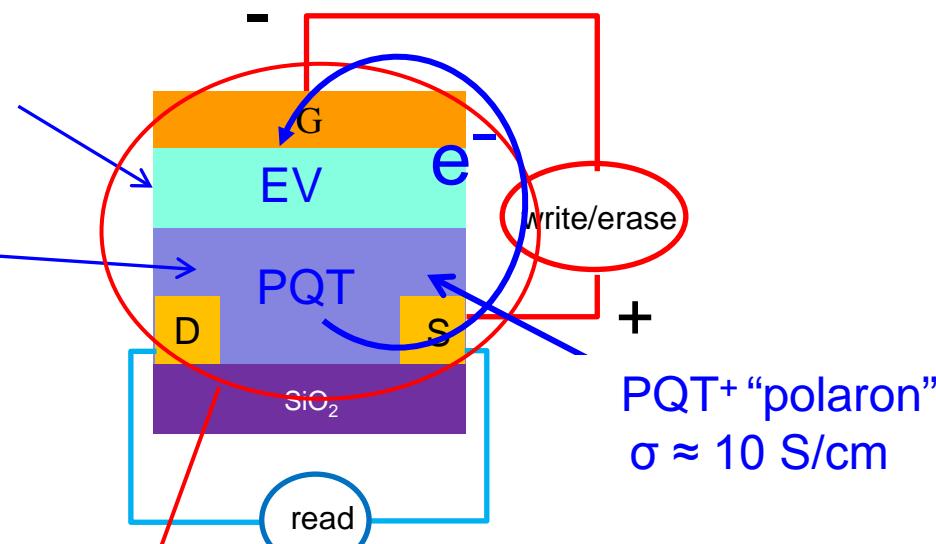


ethyl viologen ClO_4^-
polyethylene oxide

PQT from Xerox Canada



more to scale:

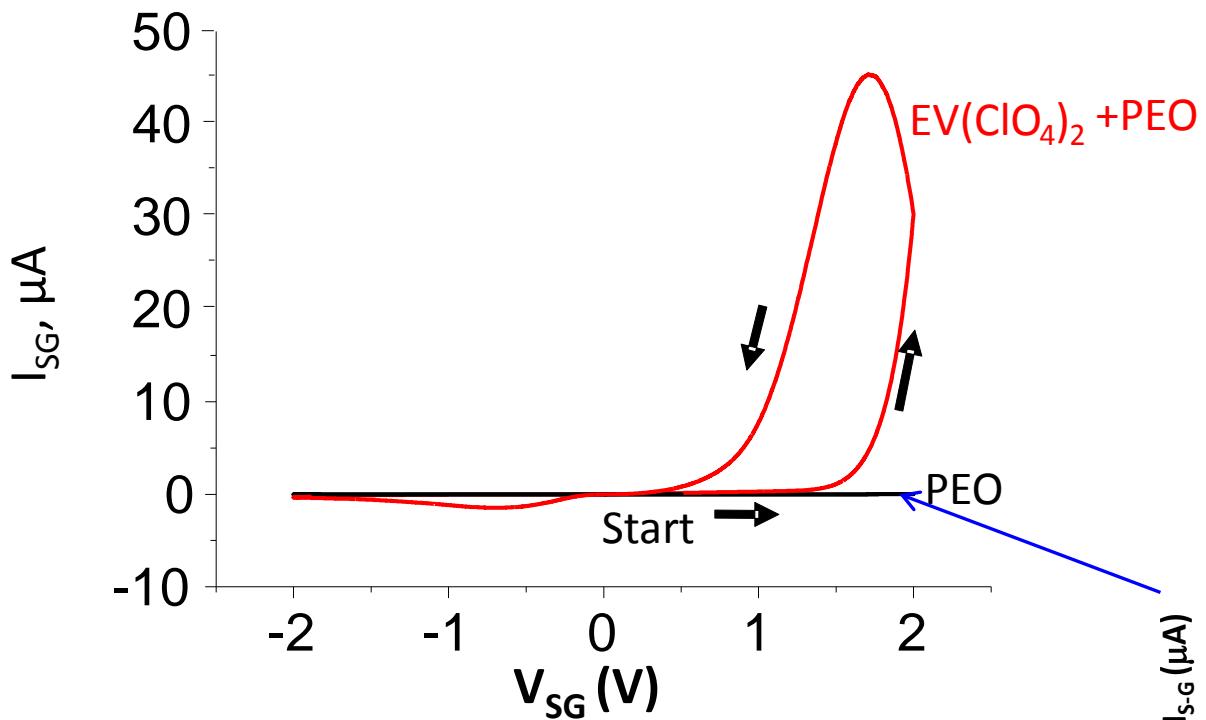
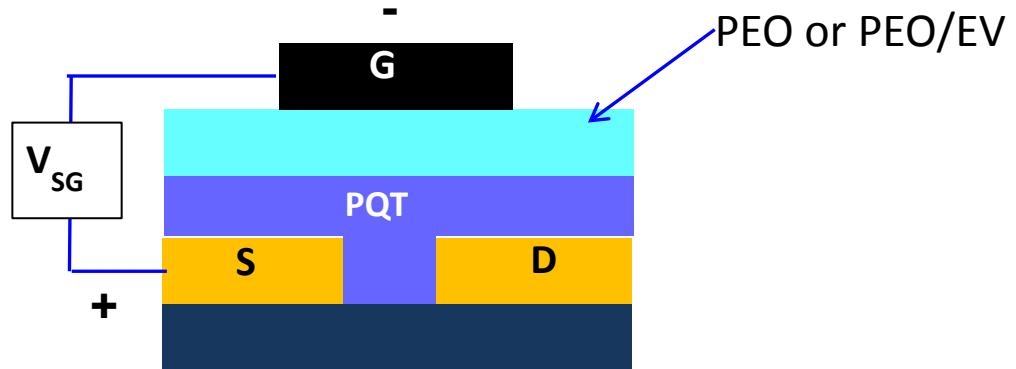


PQT⁺ “polaron”
 $\sigma \approx 10$ S/cm

1000 nm

S

25 nm

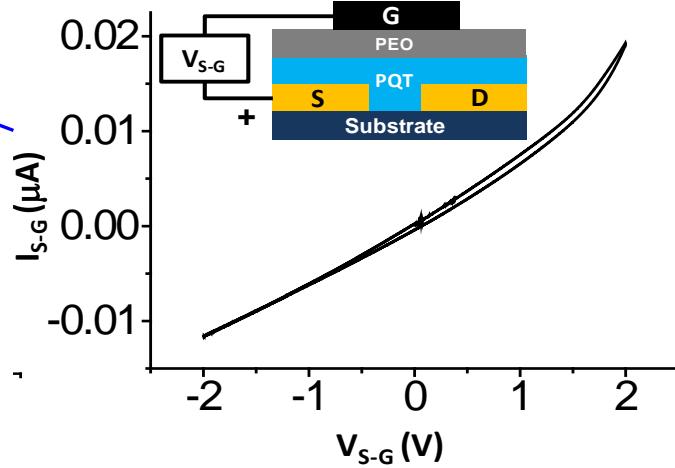
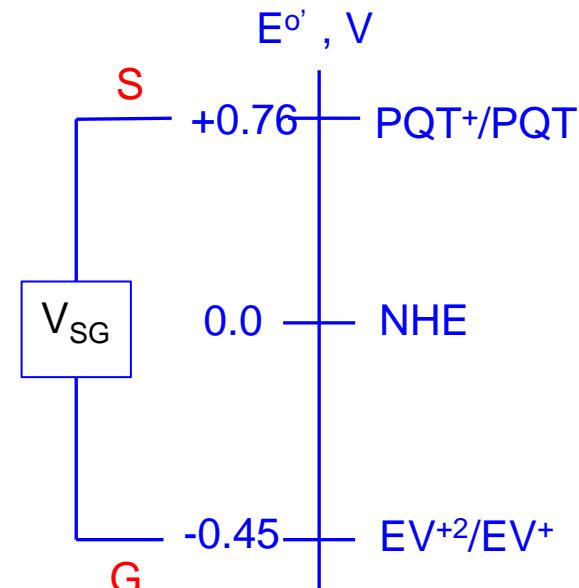


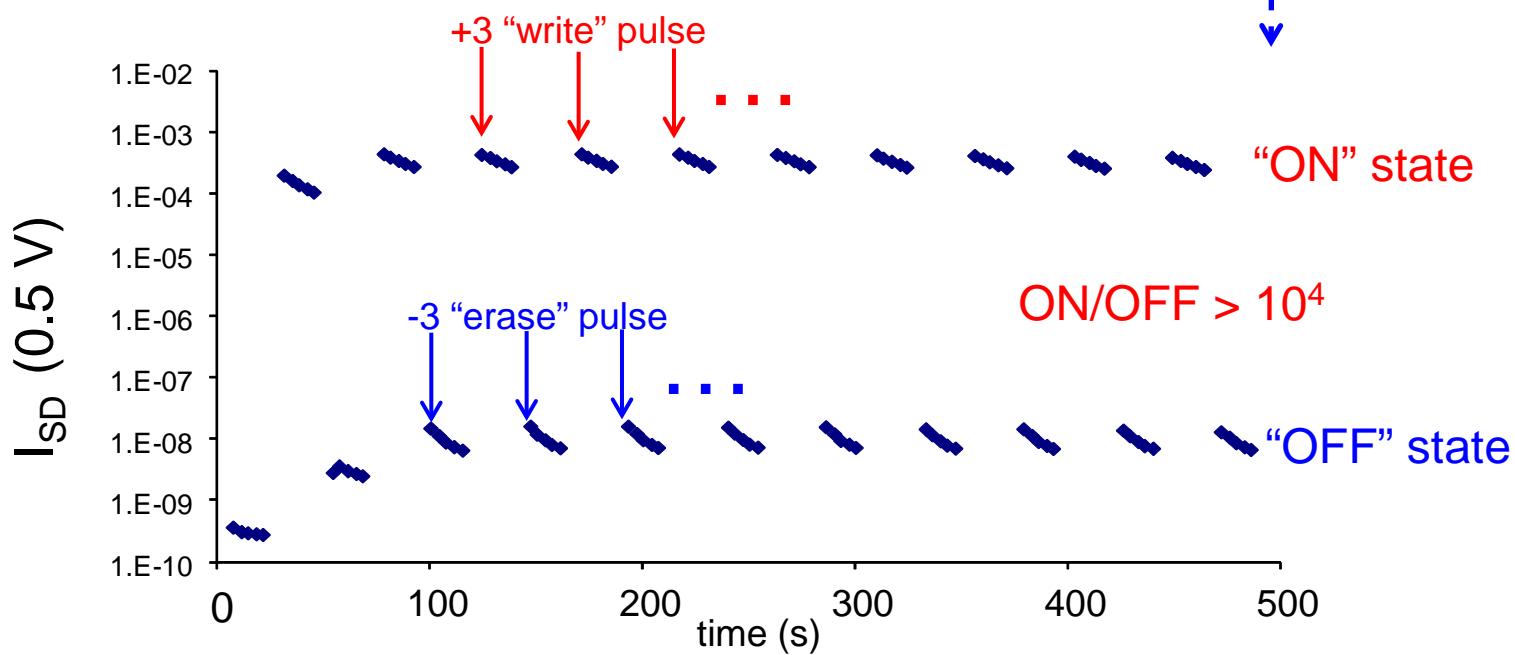
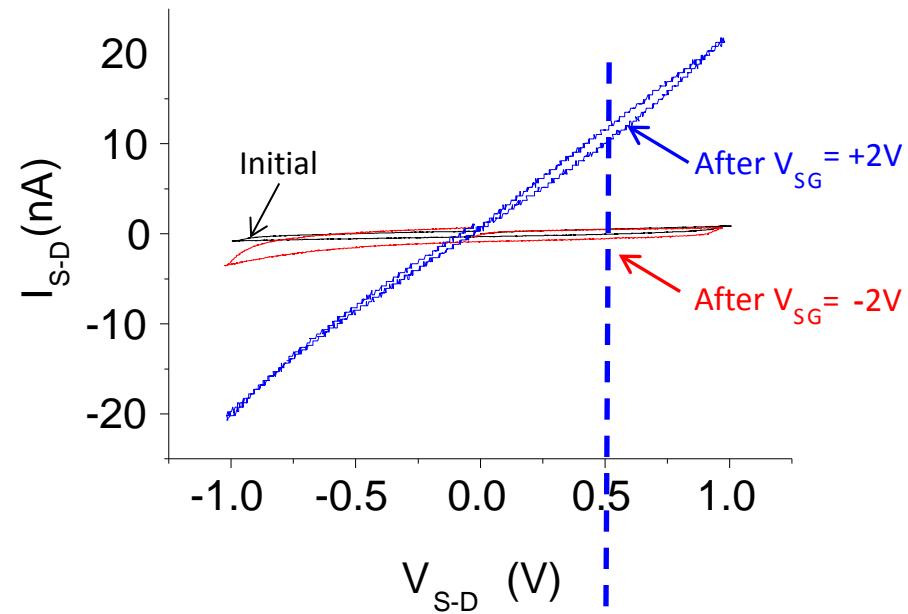
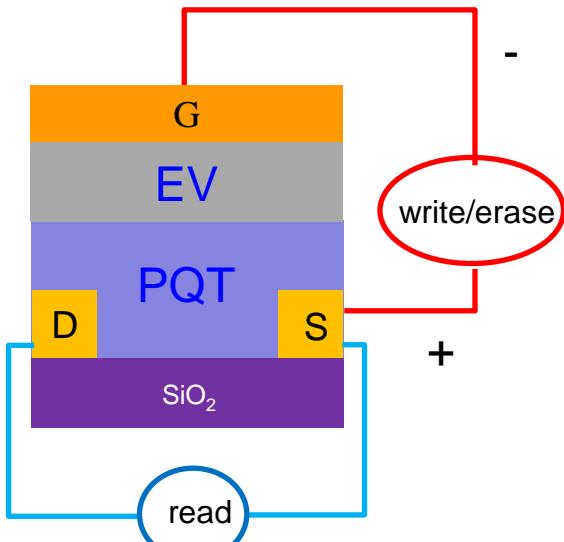
V_{SG} drives a redox reaction:

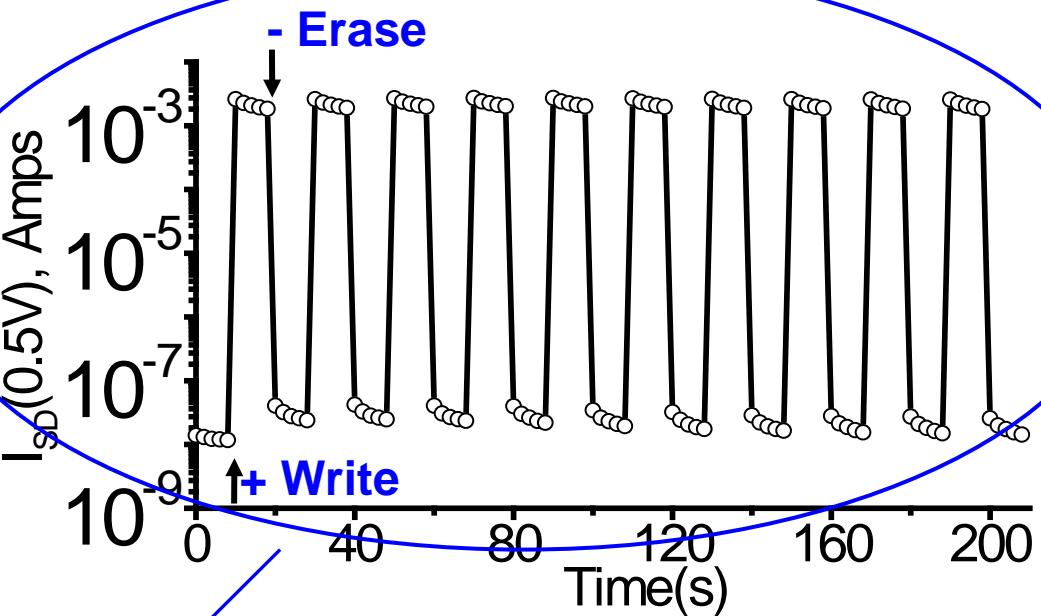
$\xrightarrow{\text{"write"}}$

$$\text{PQT} + \text{EV}^{+2} \rightarrow \text{PQT}^+ + \text{EV}^+$$

$\xleftarrow{\text{"erase"}}$



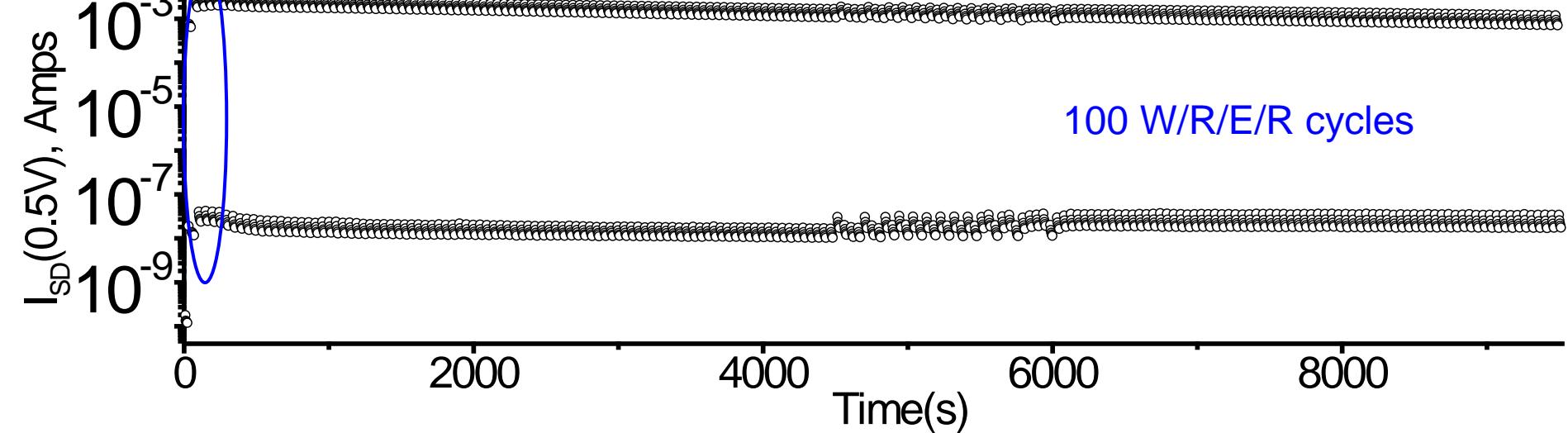




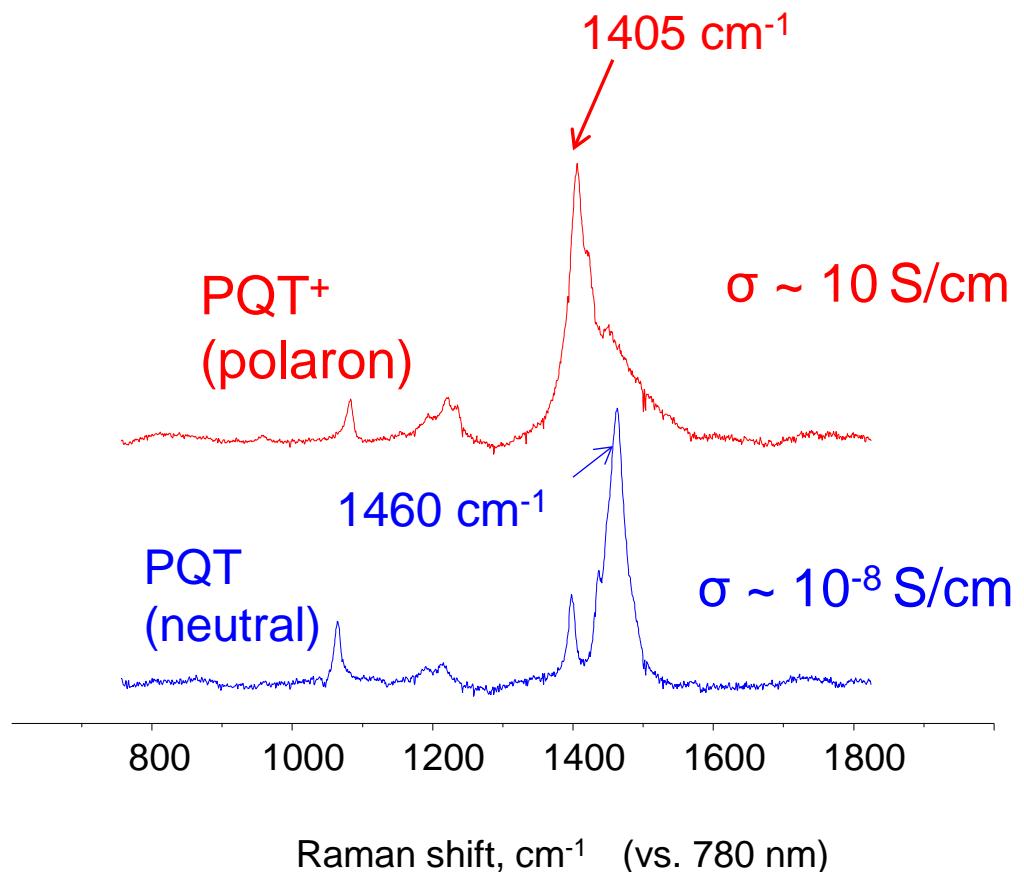
Note:

- resistance readout
- low energy W/E
- nondestructive “read”
- should scale well
- potentially longer cycle life than “flash”

100 W/R/E/R cycles

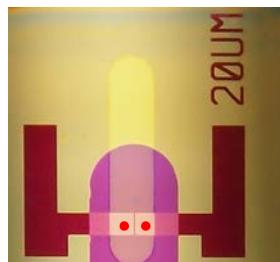


Raman spectroelectrochemistry of PQT:

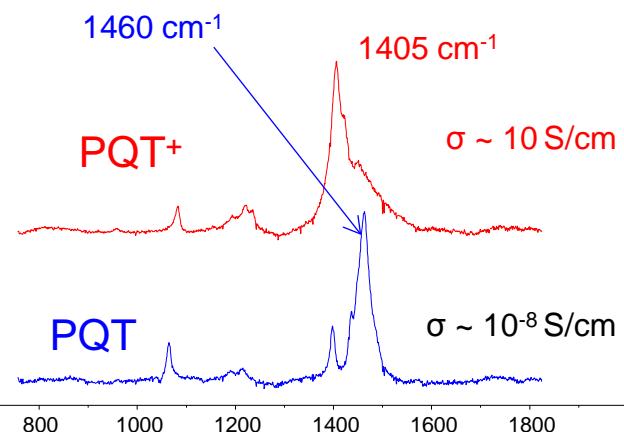
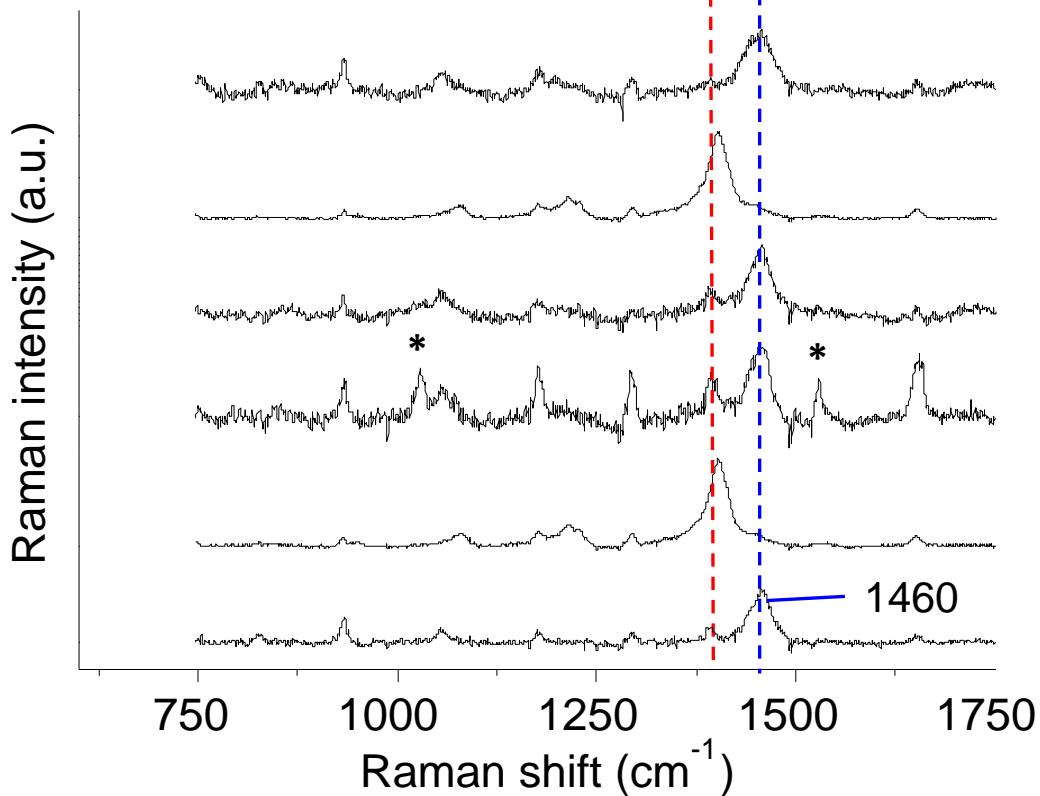
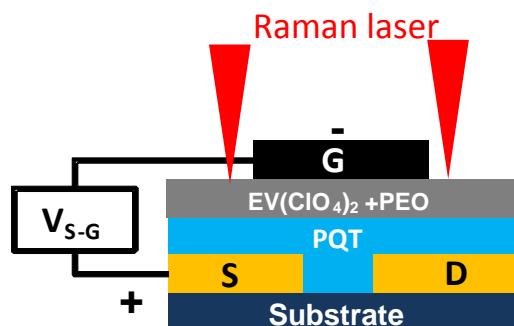


Raman permits monitoring of PQT^+ formation in *working* memory device

Solid state spectroelectrochemistry:



top view



Drain, $V_{SG} = -2$

Drain, $V_{SG} = +2$

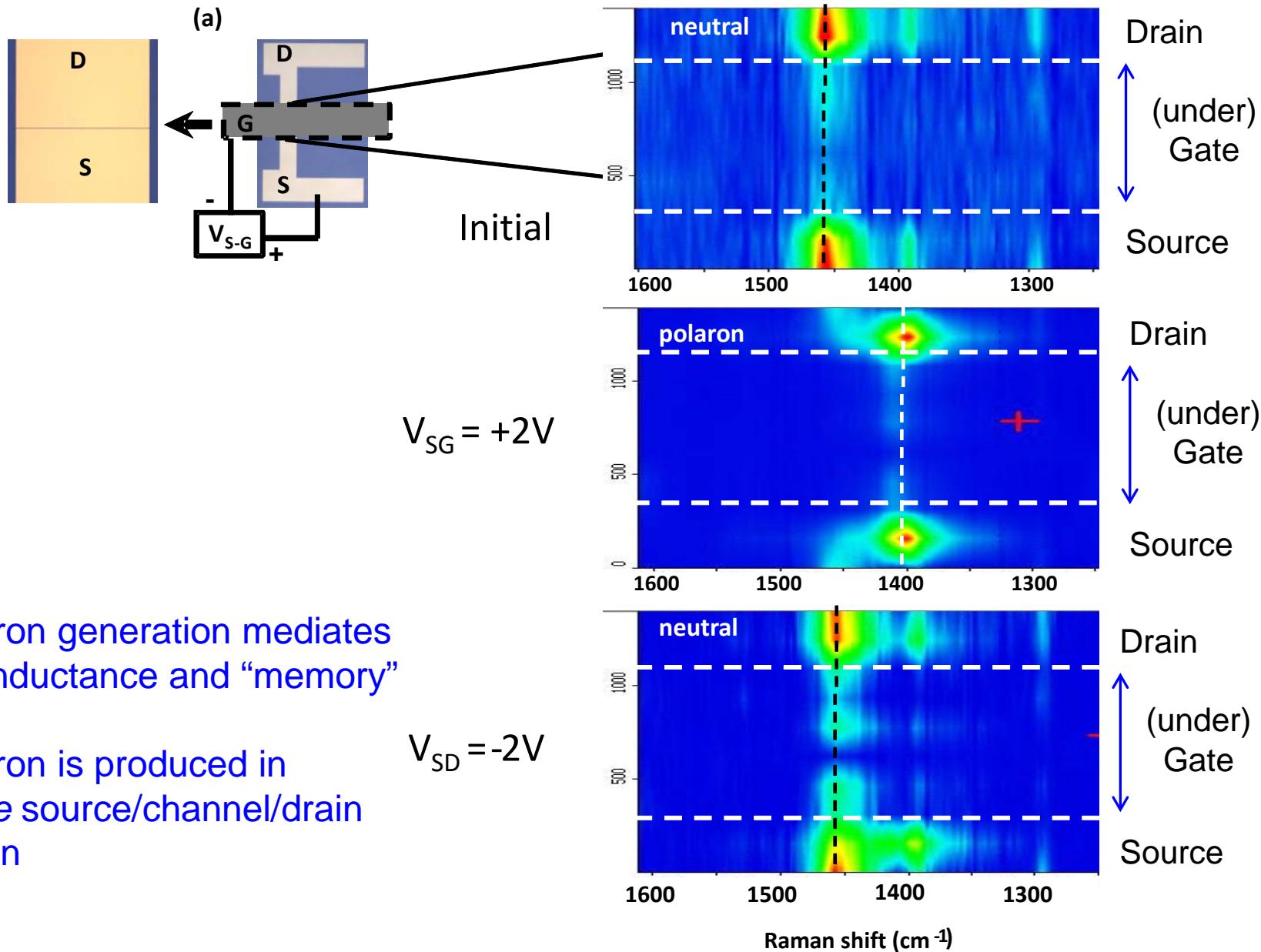
Drain, initial

Source, $V_{SG} = -2$

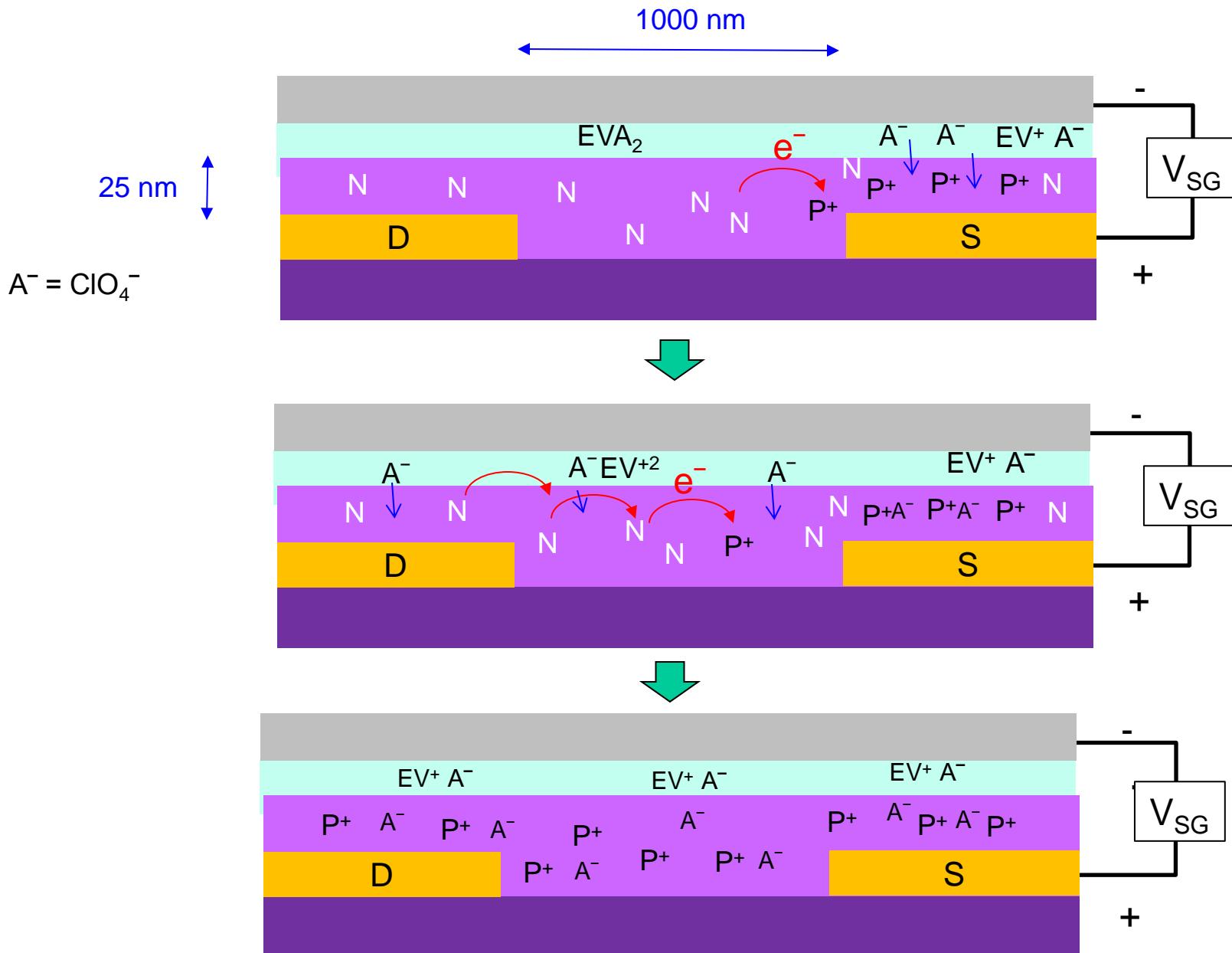
Source, $V_{SG} = +2$

Source, initial

V_{SG} pulses are
“switching” polymer
between high and low
conductance states

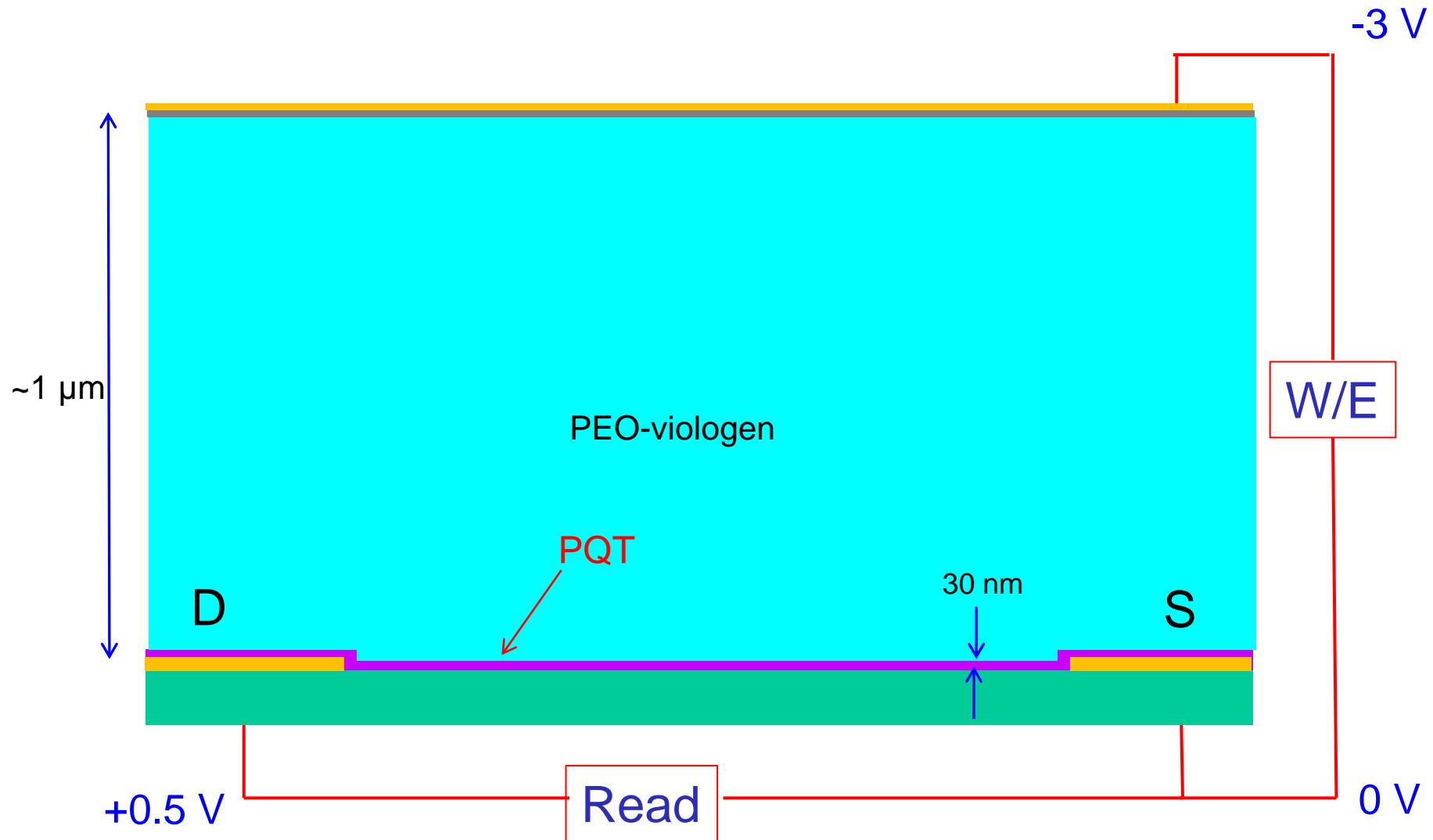


- polaron generation mediates conductance and “memory”
- polaron is produced in *entire* source/channel/drain region

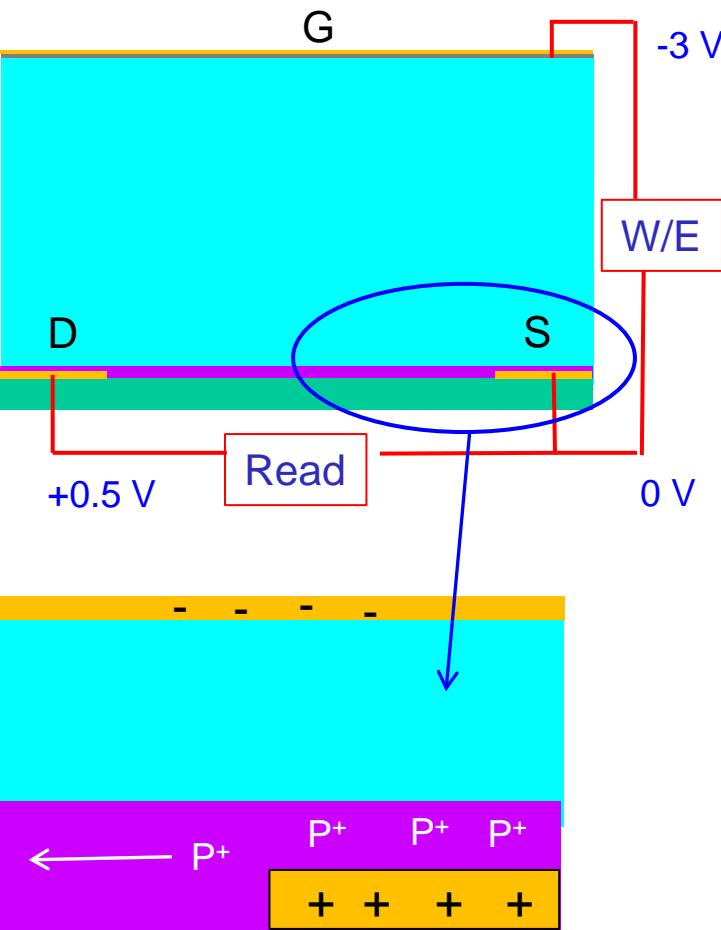


Polythiophene is not only a redox polymer but also a conducting polymer, so electron transport “laterally” is quite efficient.

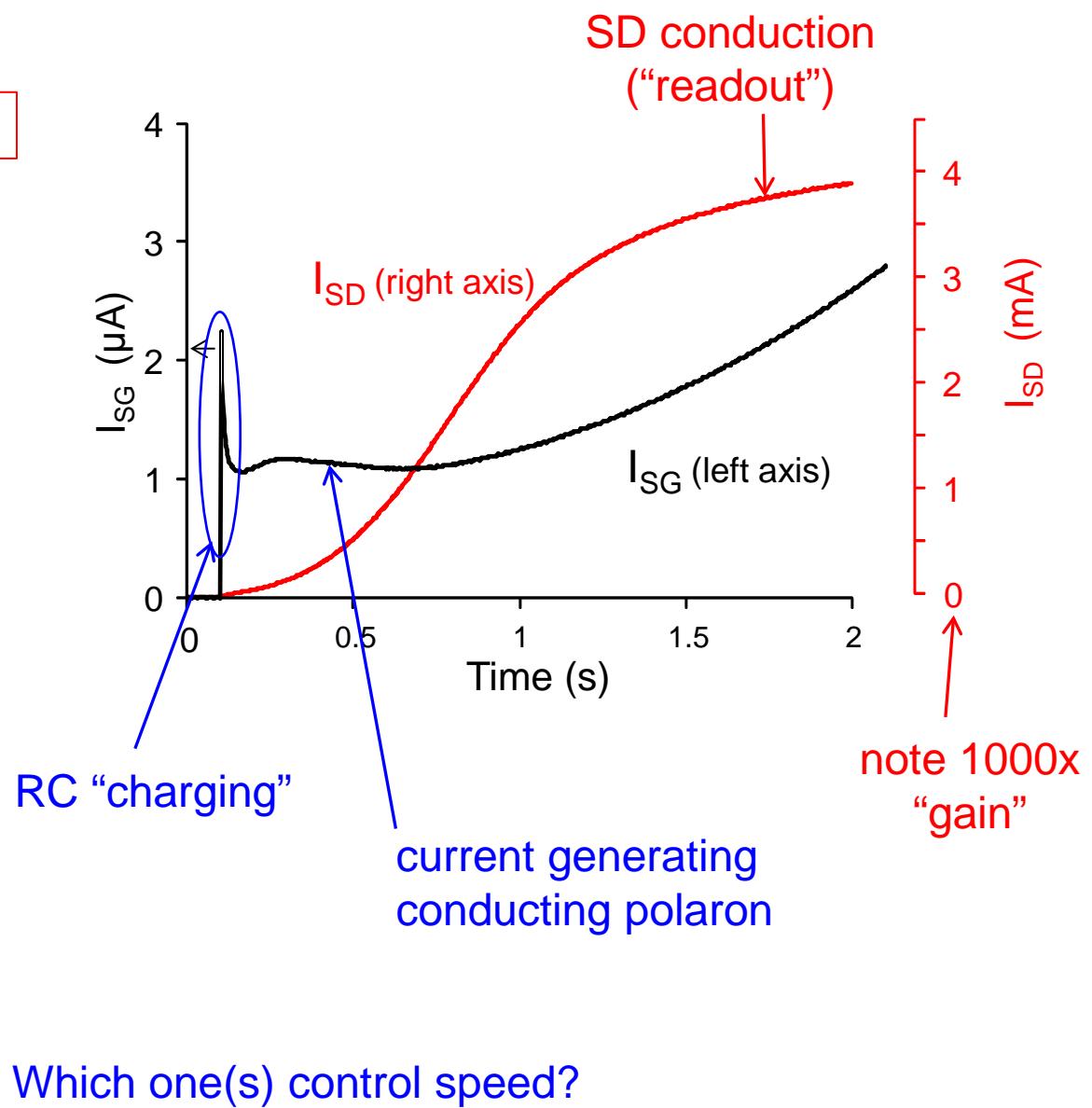
A closer look at dynamics, using a circuit equivalent to a bi-potentiostat:



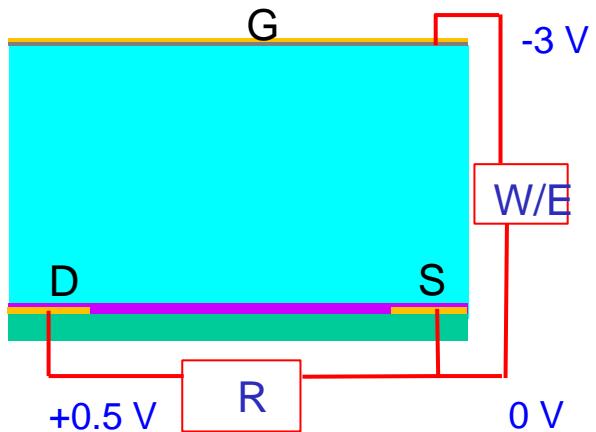
Now, monitor S-D current during $+3 \text{ V}$ S-G “write” pulse



- charging (RC)
- P^+ generation
- P^+ propagation



An interesting effect
of atmosphere:



RC charging?
ionic mobility?
propagation speed?

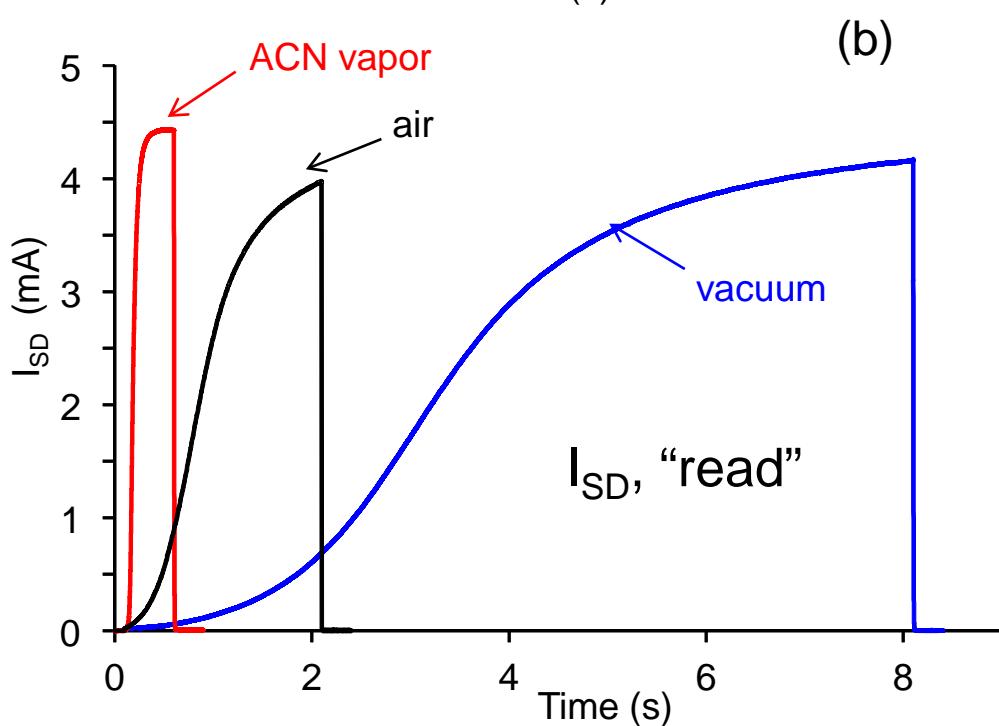
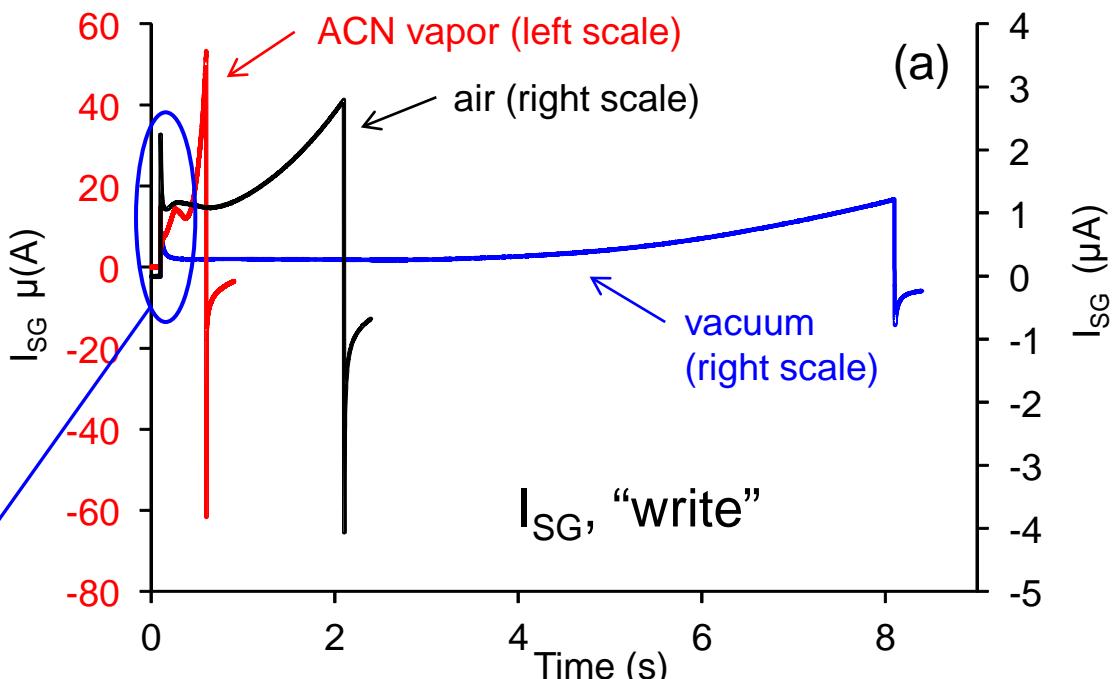
RC (msec)

Ambient 0.18

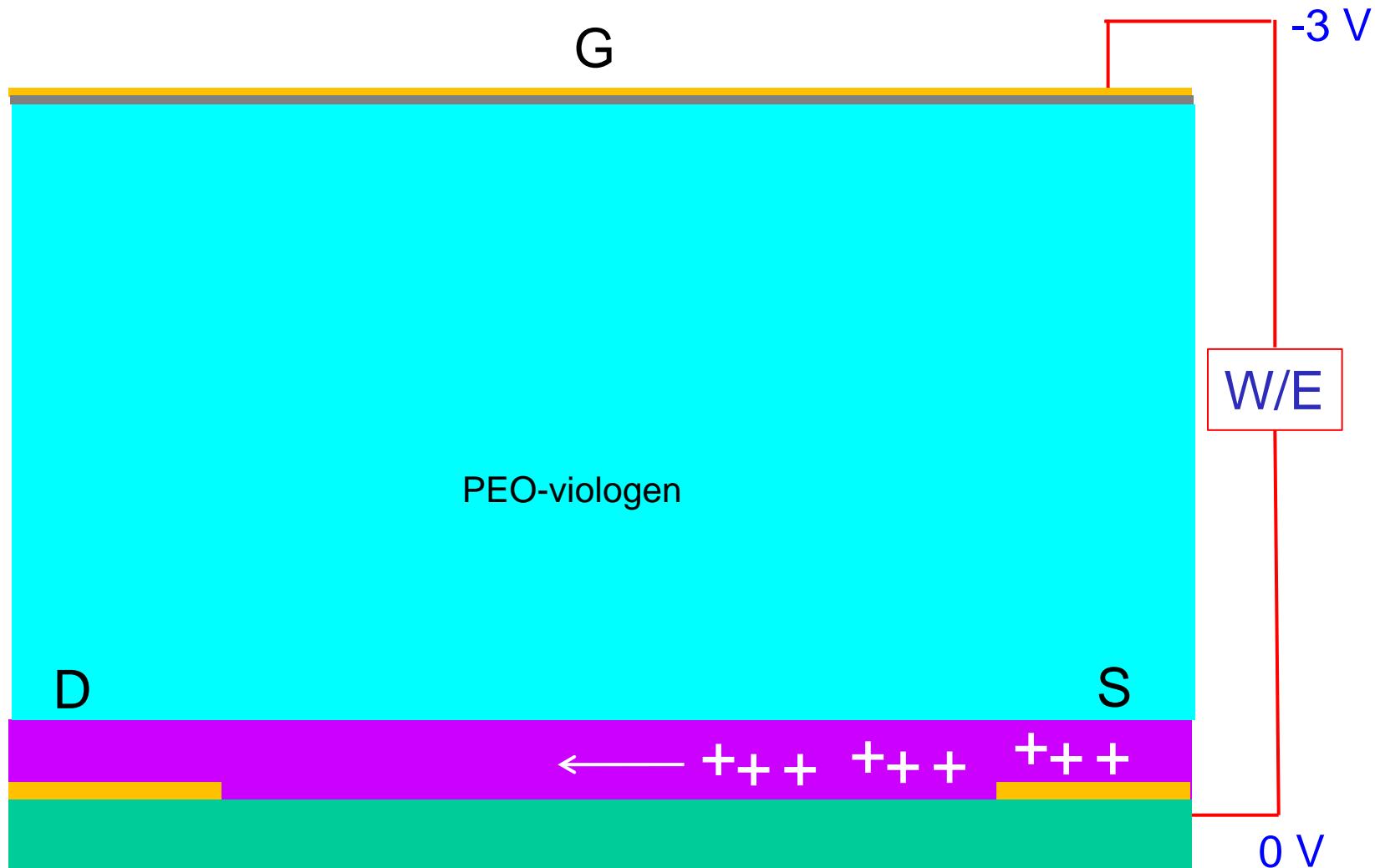
Vacuum 0.037

ACN vapor 0.081

all < 1 msec, can't be the
problem

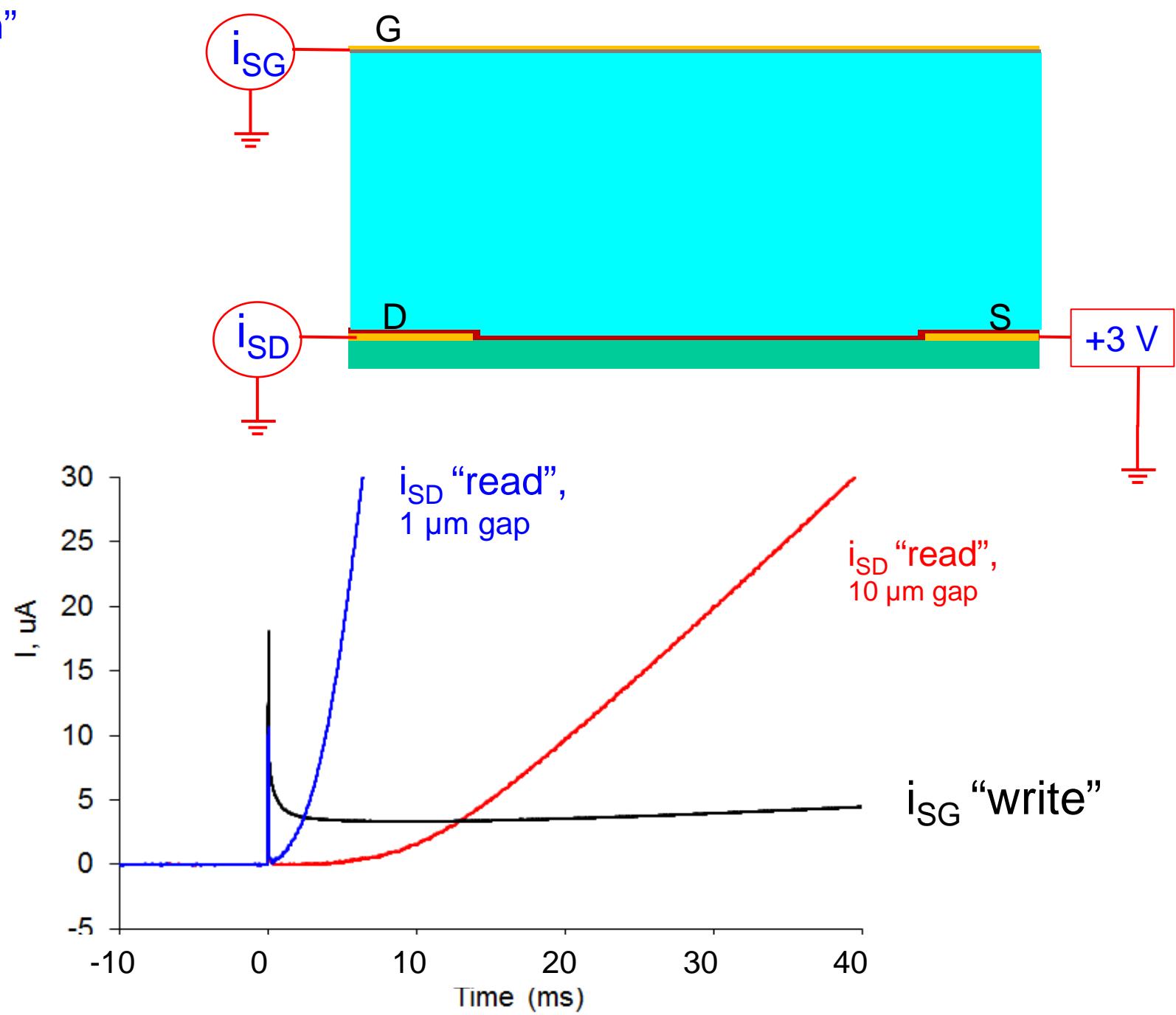


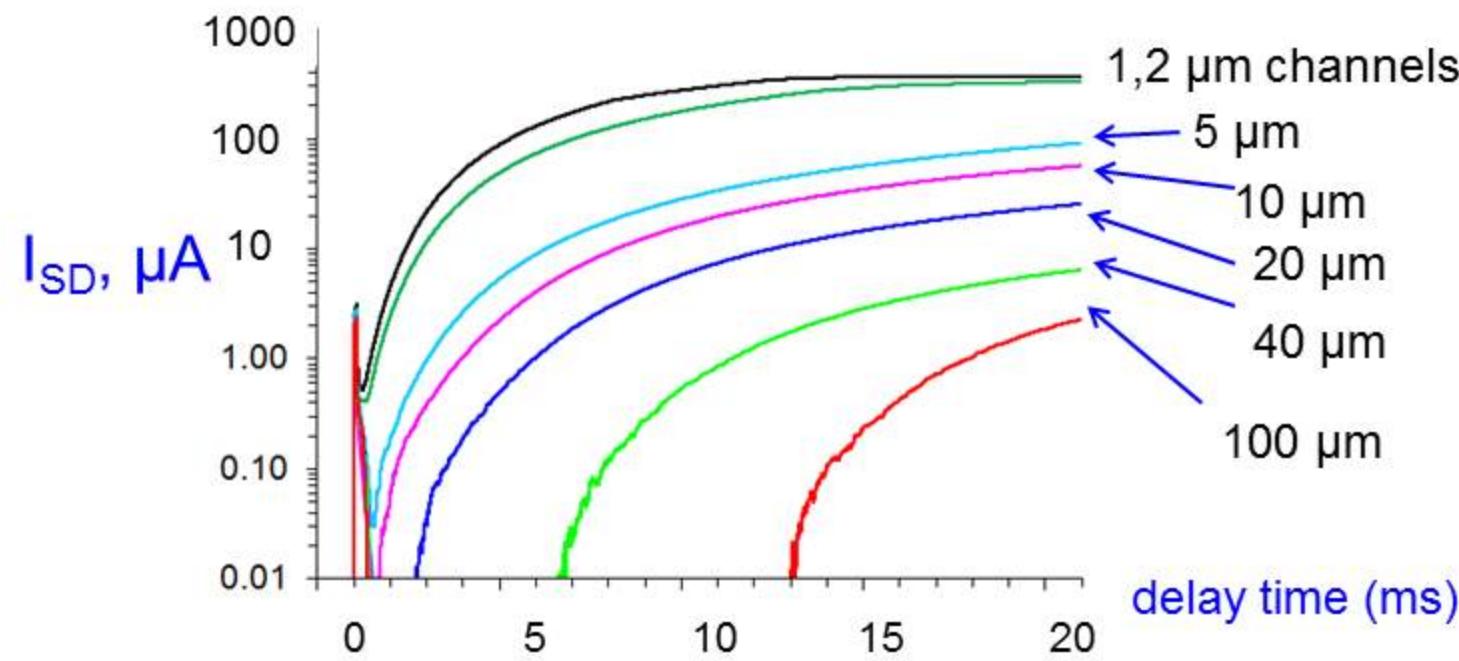
Propagation of polarons into channel:

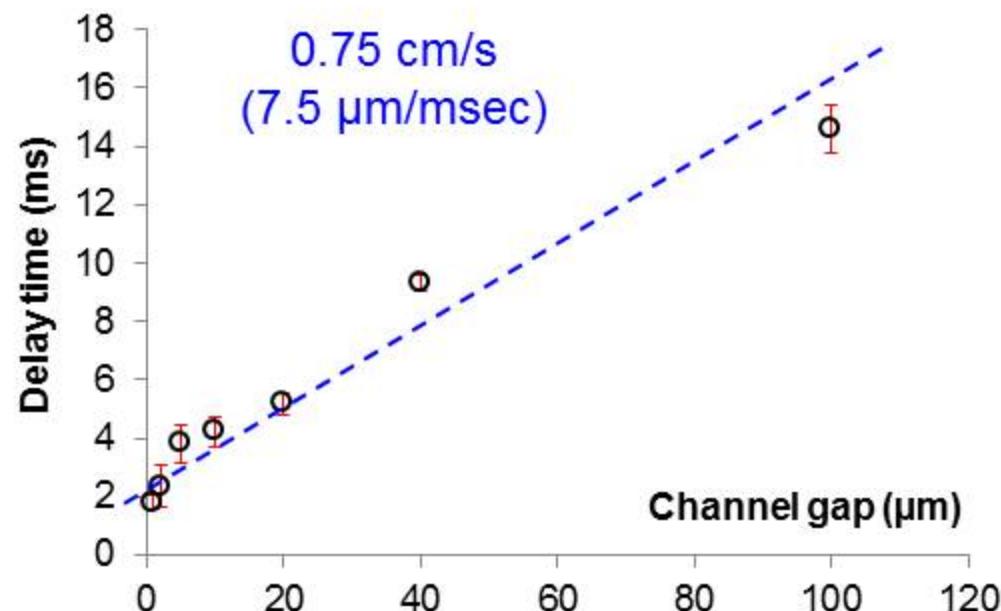
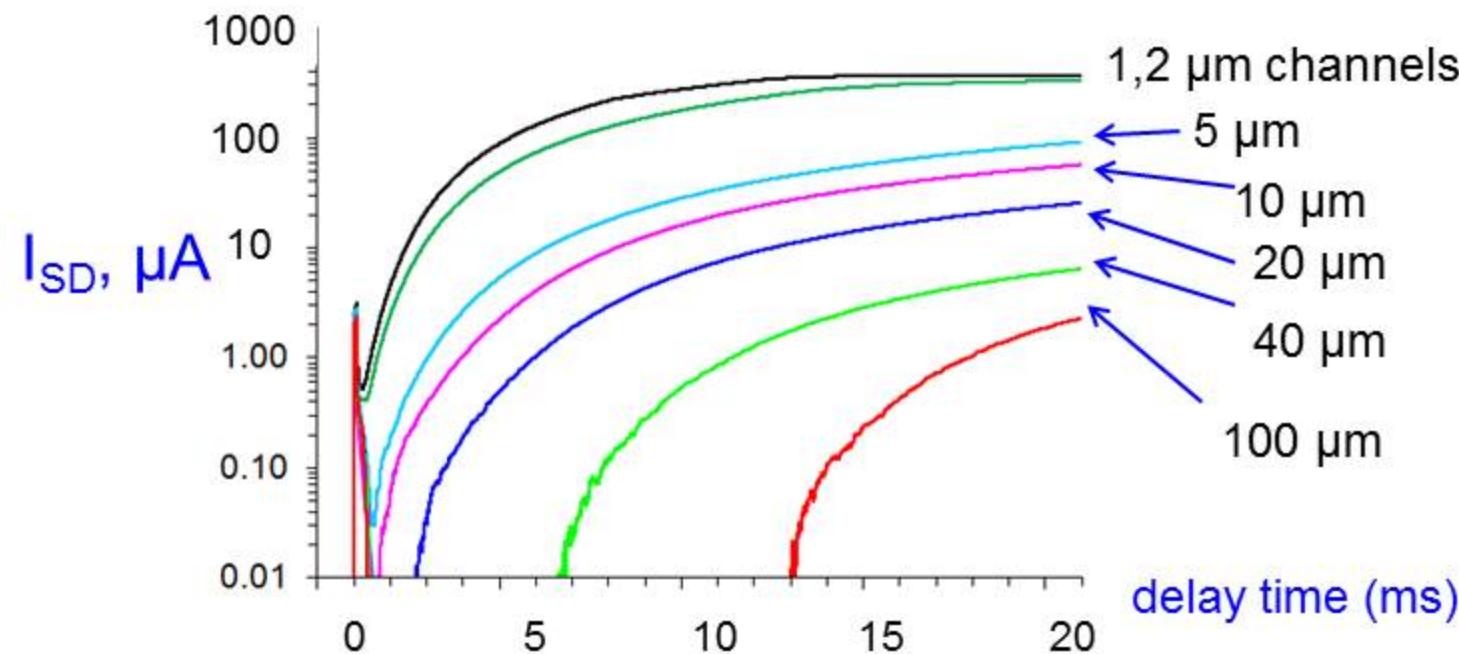


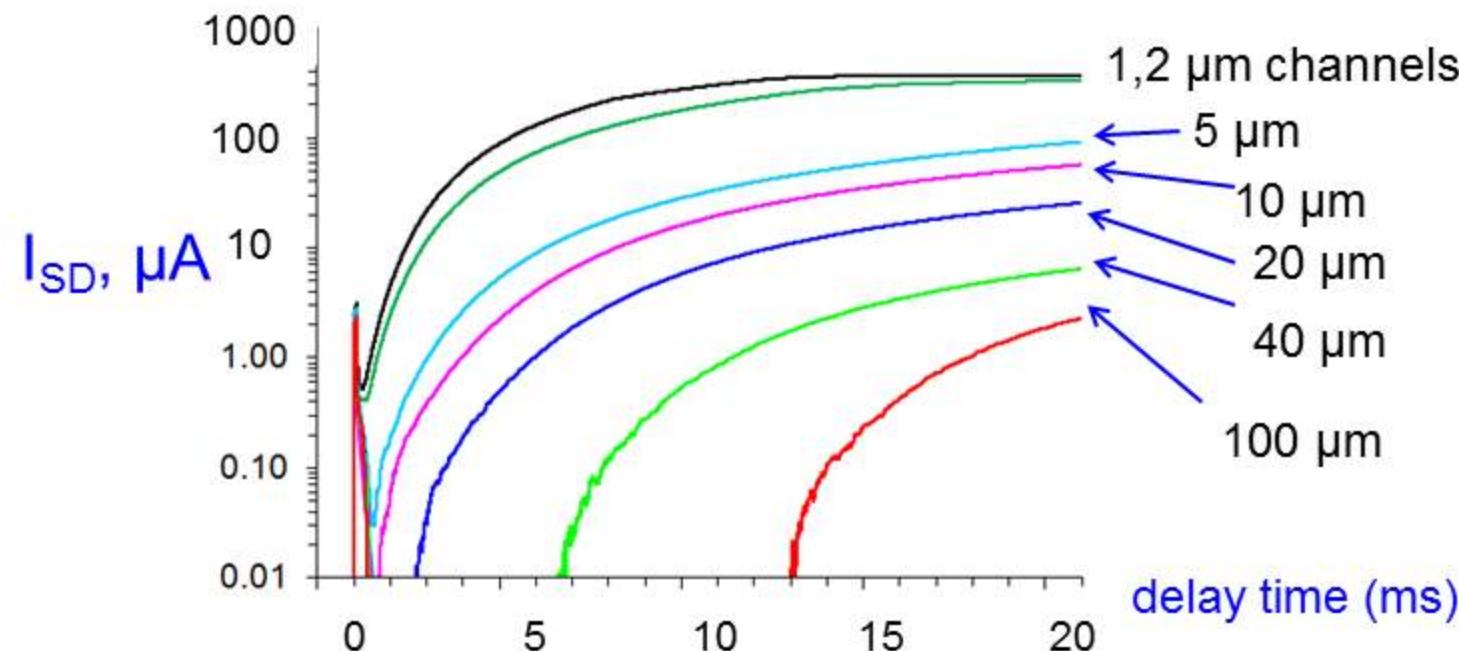
Propagation is essential, could it be the slow step?

“Propagation” experiment:

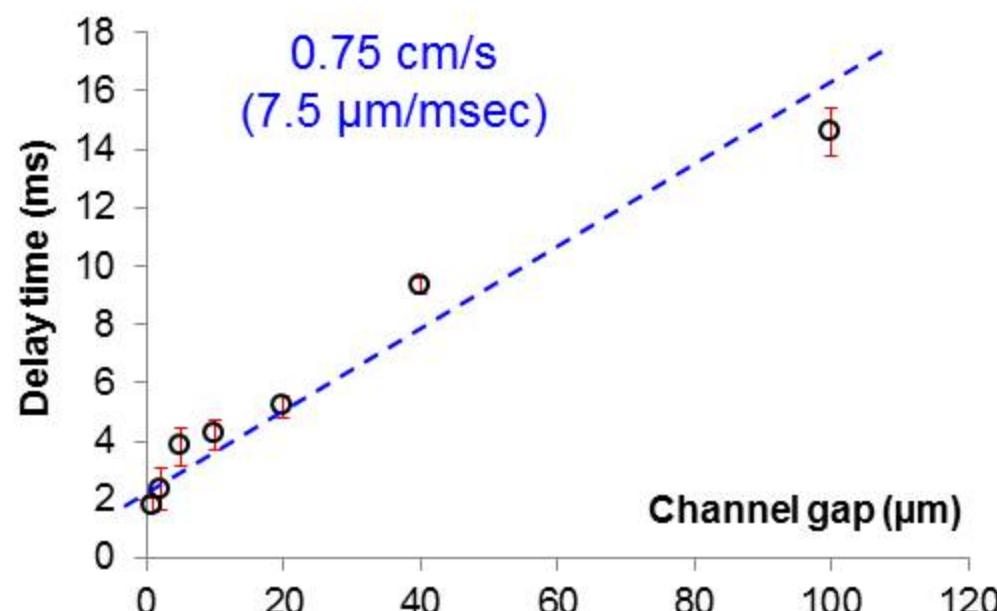






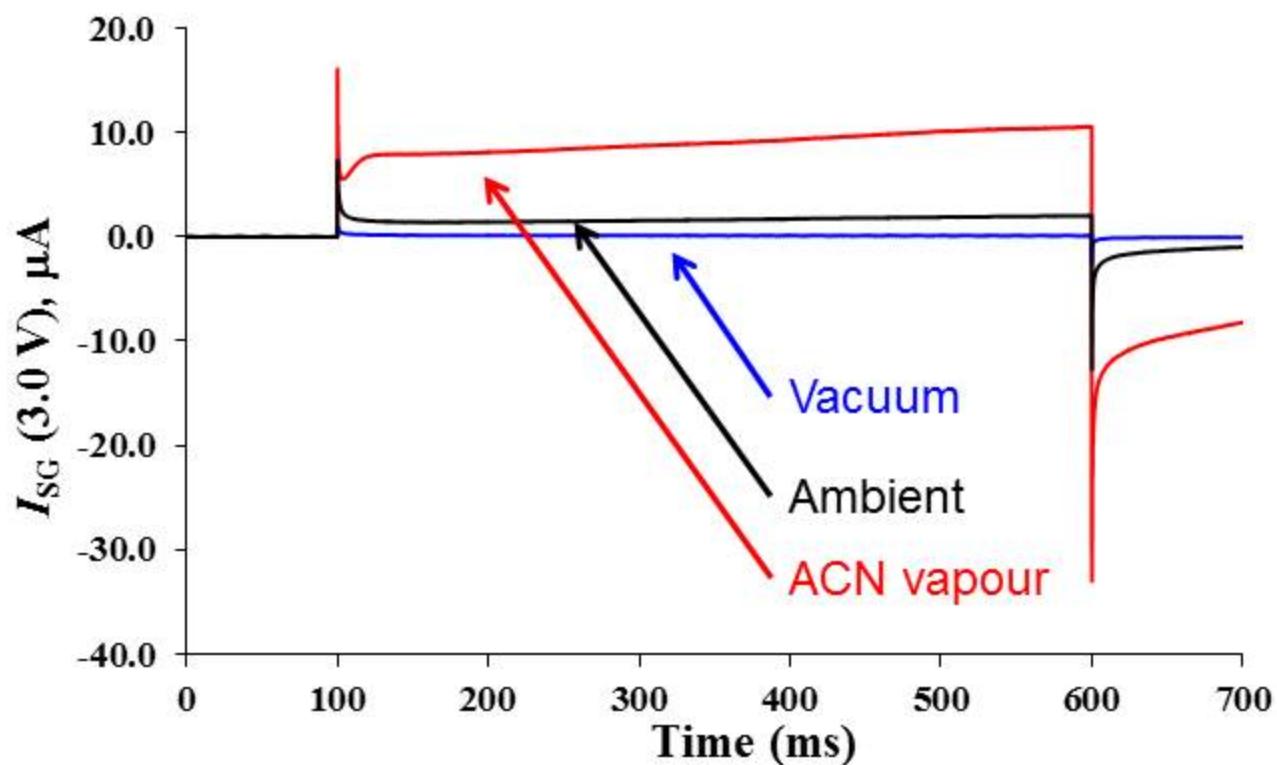


for our 1-5 μm channels,
propagation delay is
only a few msec.
Cannot explain “write”
times of ~ 1 sec



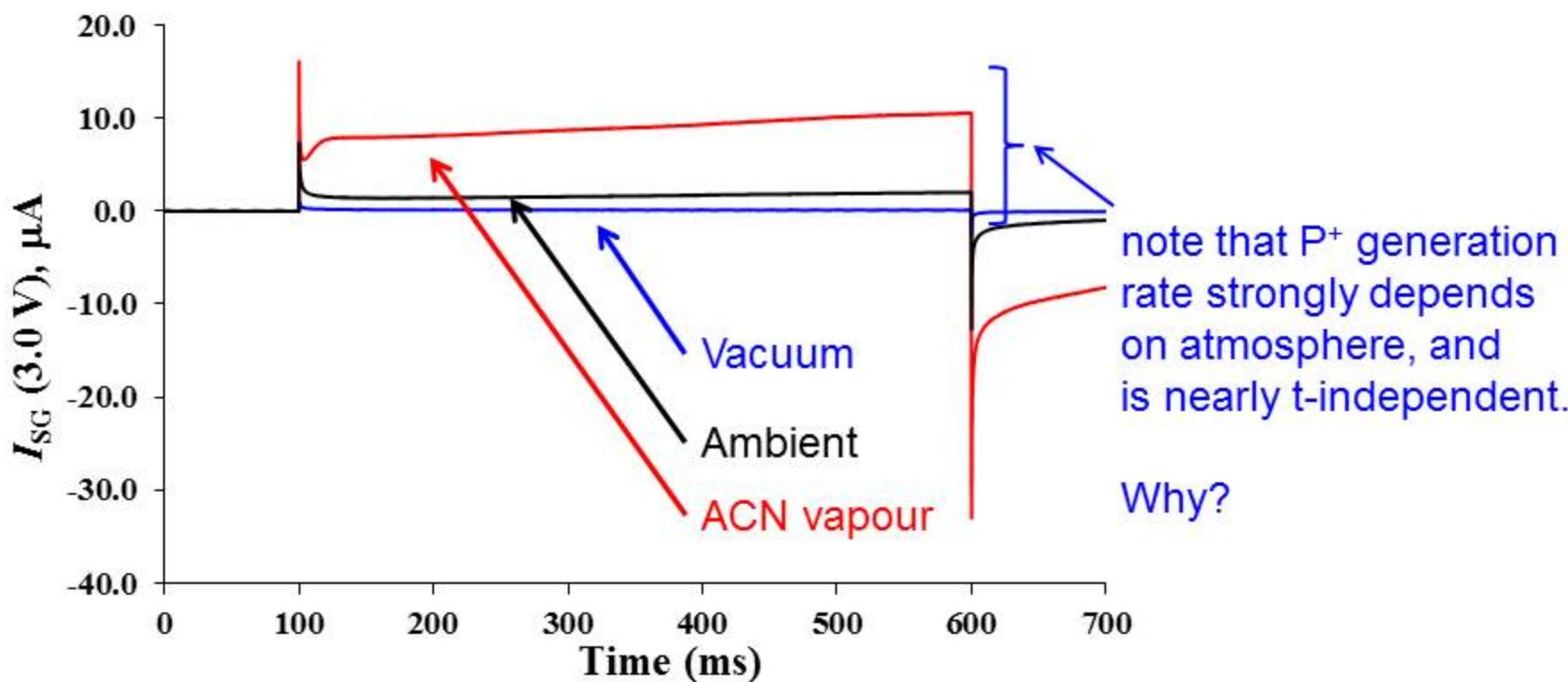
I_{SG} "write" current:

	RC (msec)	R (Ω)
Ambient	0.18	250 k
Vacuum	0.037	240 k
ACN vapor	0.081	60 k

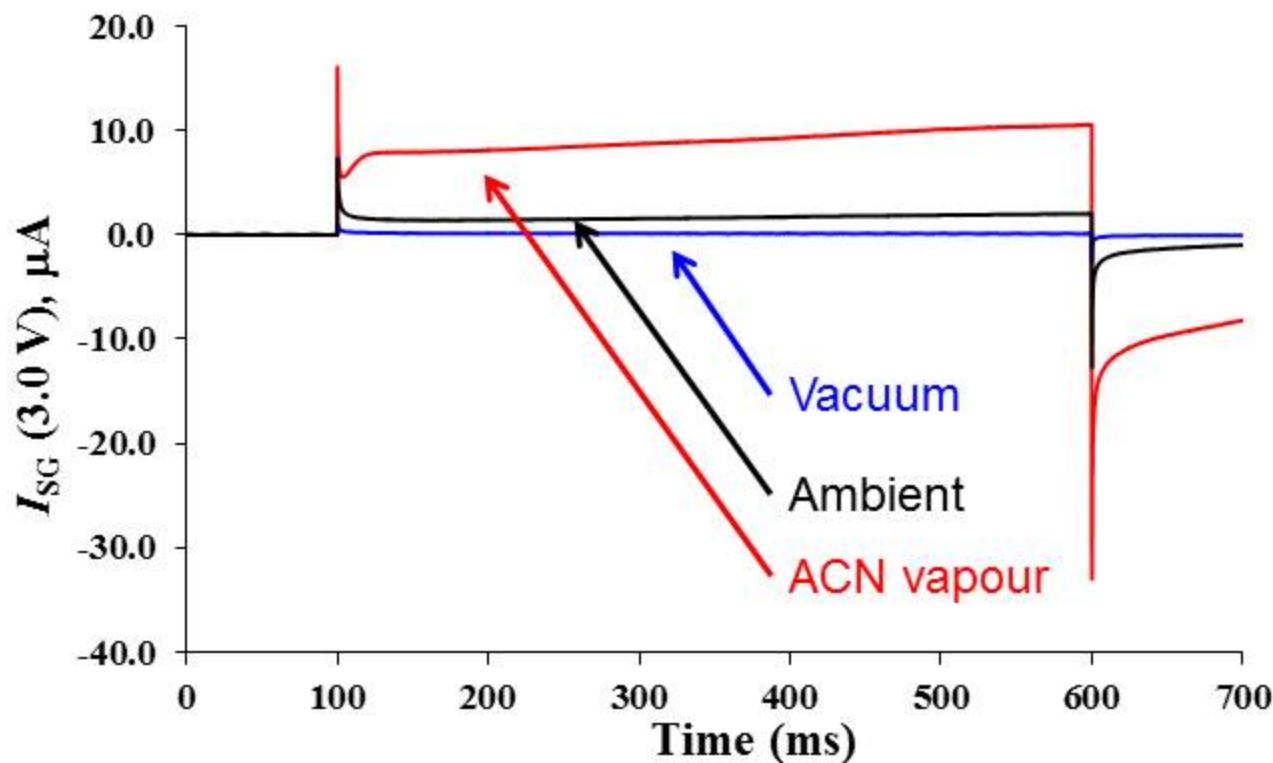


I_{SG} "write" current:

	RC (msec)	R (Ω)
Ambient	0.18	250 k
Vacuum	0.037	240 k
ACN vapor	0.081	60 k



	RC (msec)	R	iR for i= 5 uA	
Ambient	0.18	250 kΩ	1.25 V	
Vacuum	0.037	240 kΩ	1.21 V	major ohmic losses in electrolyte !
ACN vapor	0.081	60 kΩ	0.30 V	

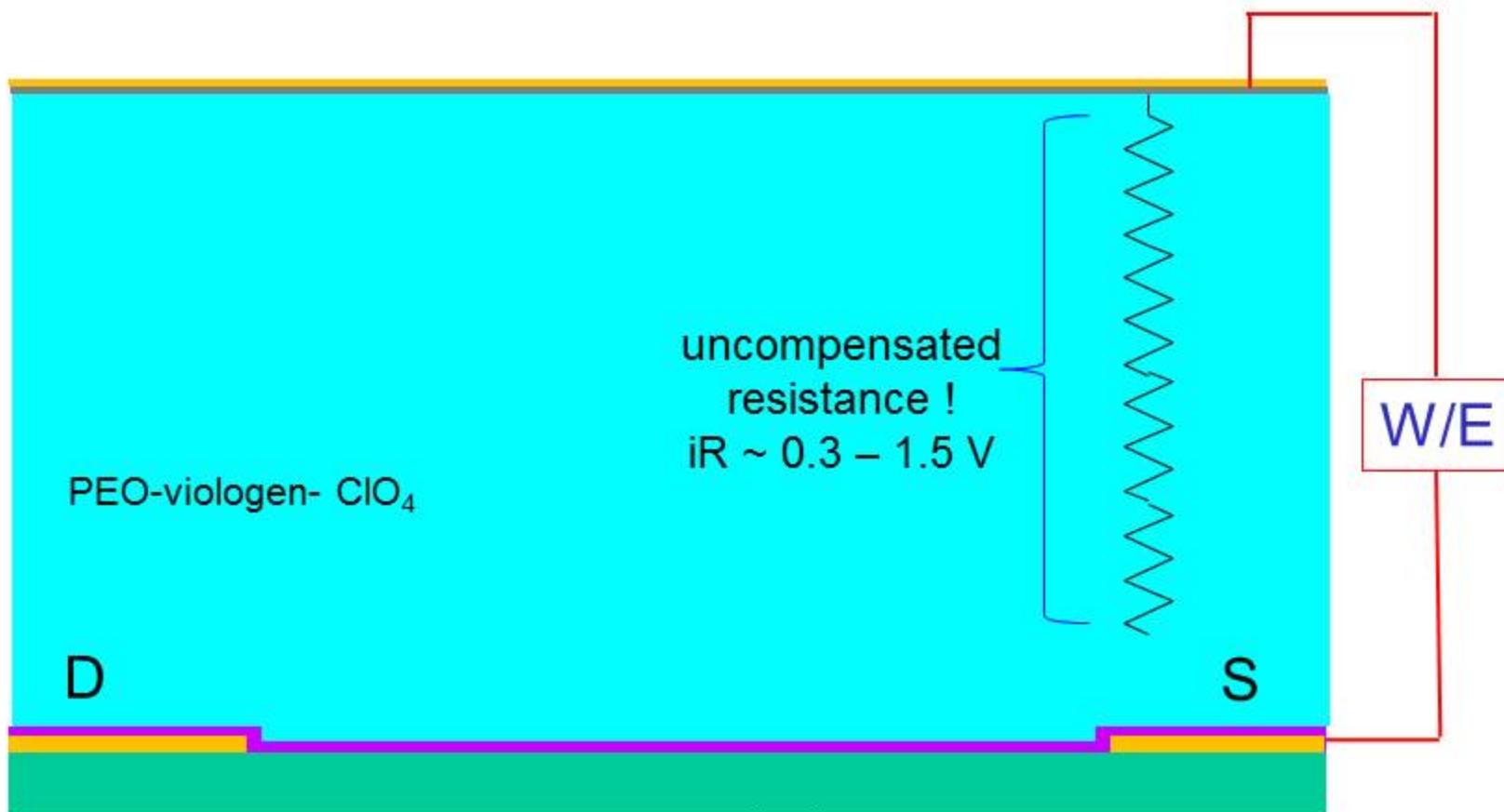


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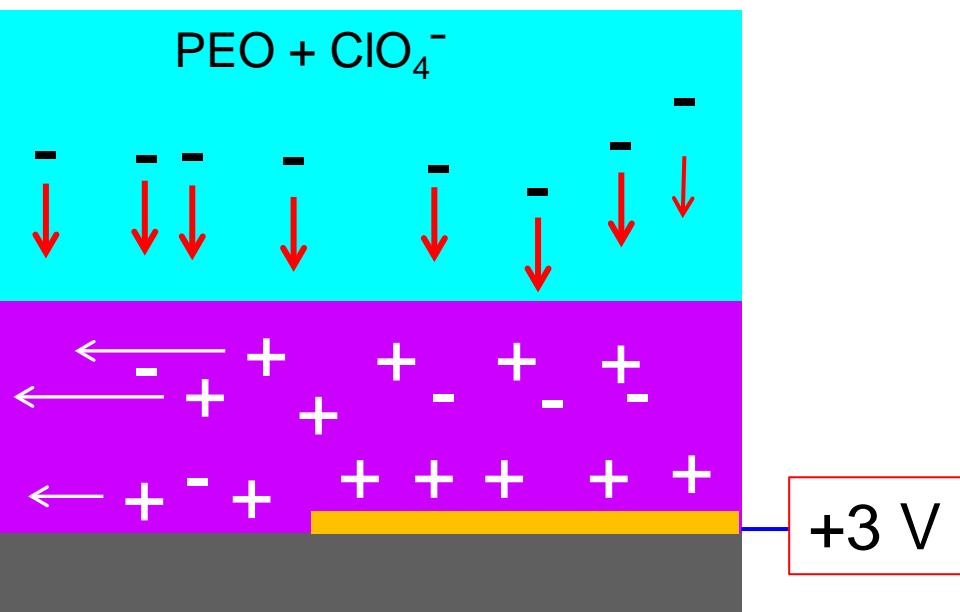


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Ambient	0.18	250 kΩ	1.25 V
Vacuum	0.037	240 kΩ	1.21 V
ACN vapor	0.081	60 kΩ	0.30 V

} major ohmic losses in electrolyte !



close-up of S electrode:

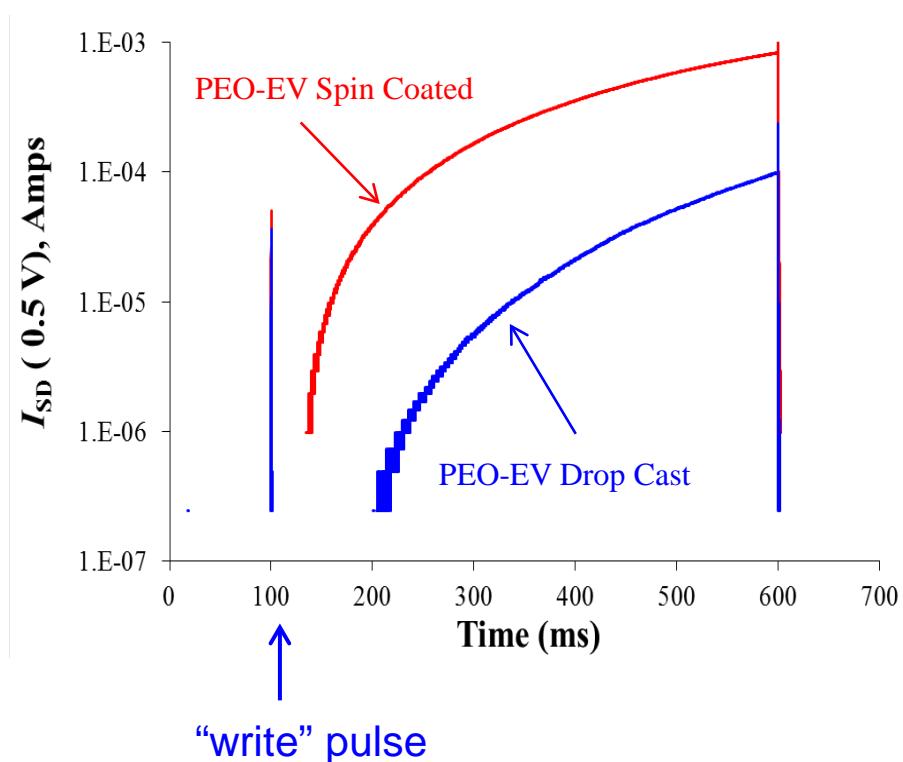
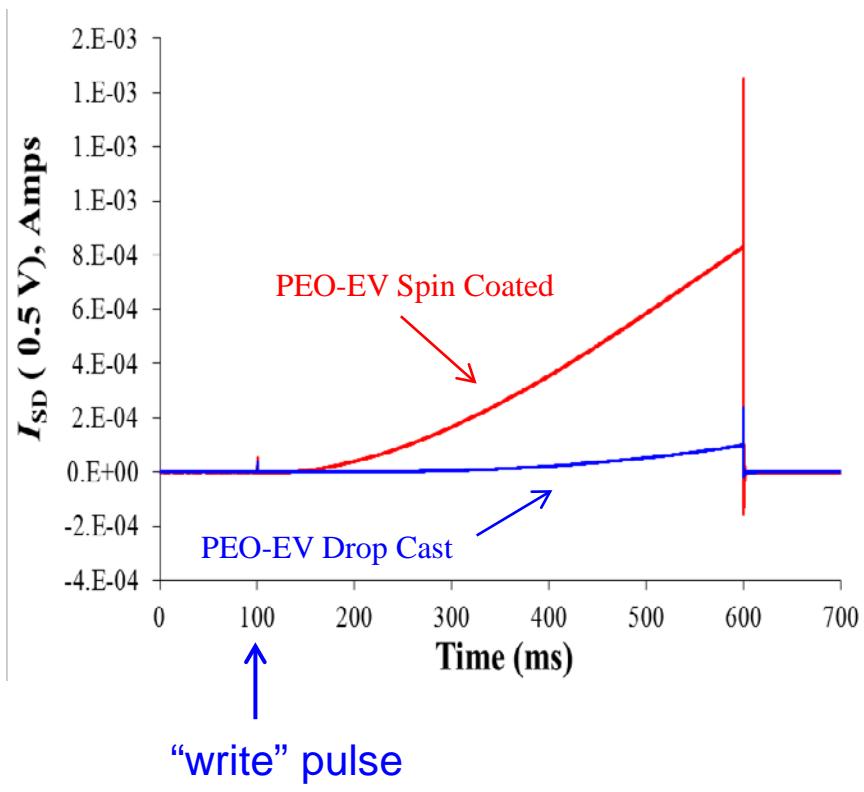
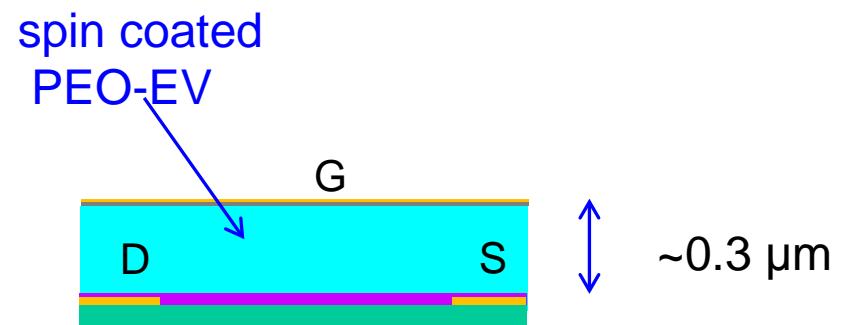
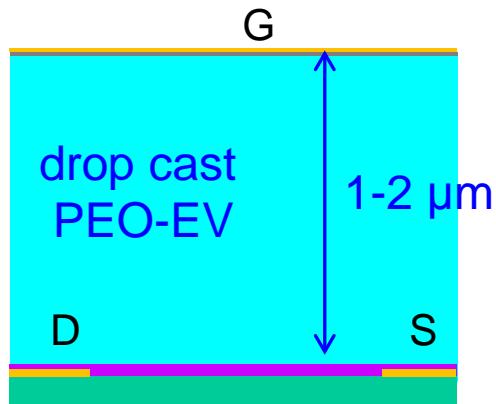


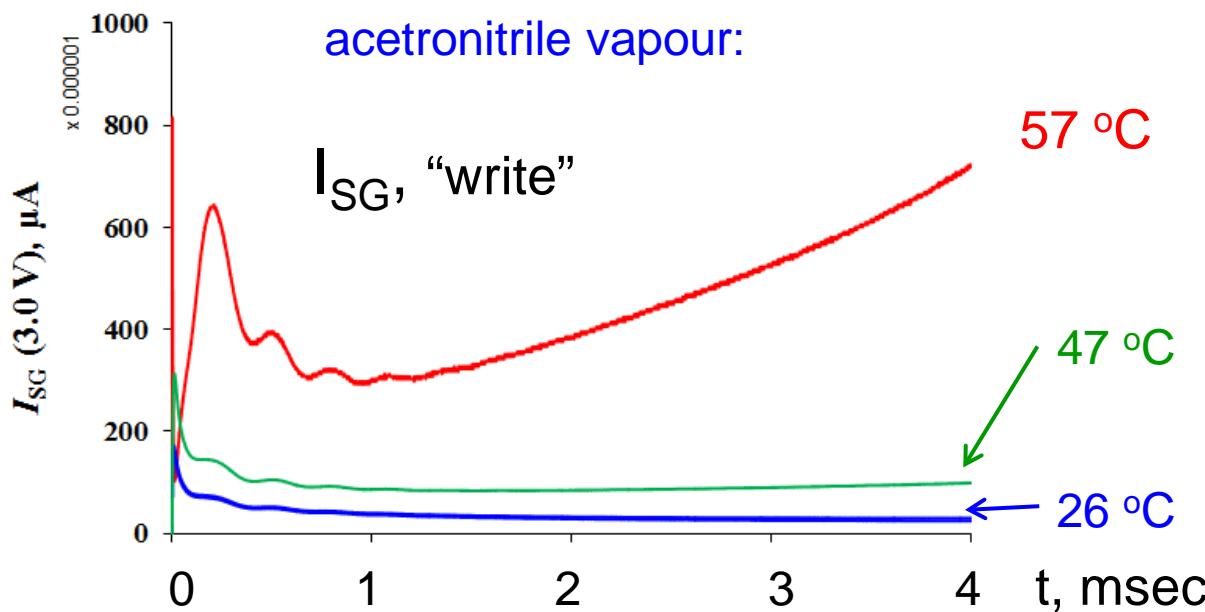
- “W” pulse generates polarons
- anions move toward polarons, compensate + charge
- polarons “fill” the Source region
- polarons propagate throughout PQT layer, anions continue to migrate from PEO

the bad news: iR losses in PEO layer reduce “write” current, thus requiring ~ 500 msec to “fill” source region.

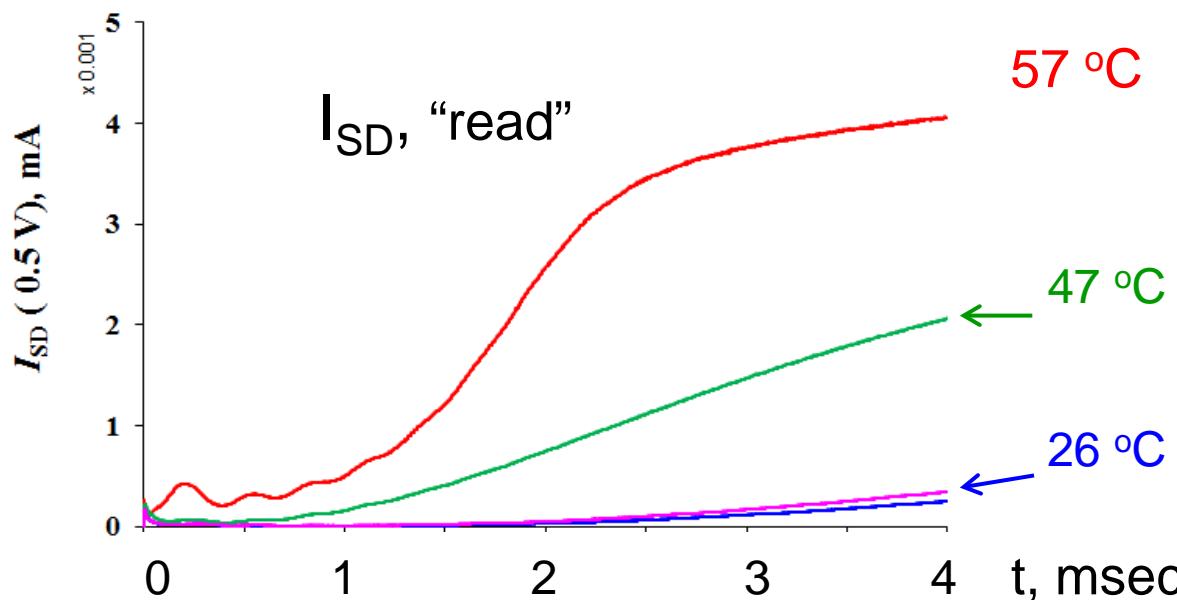
the good news: there are solid electrolytes with conductivity >1000x higher than PEO-ClO₄, which should reduce “write” time to 10 - 100 μ sec. We can also make the electrolyte at least 10x thinner

Some progress:



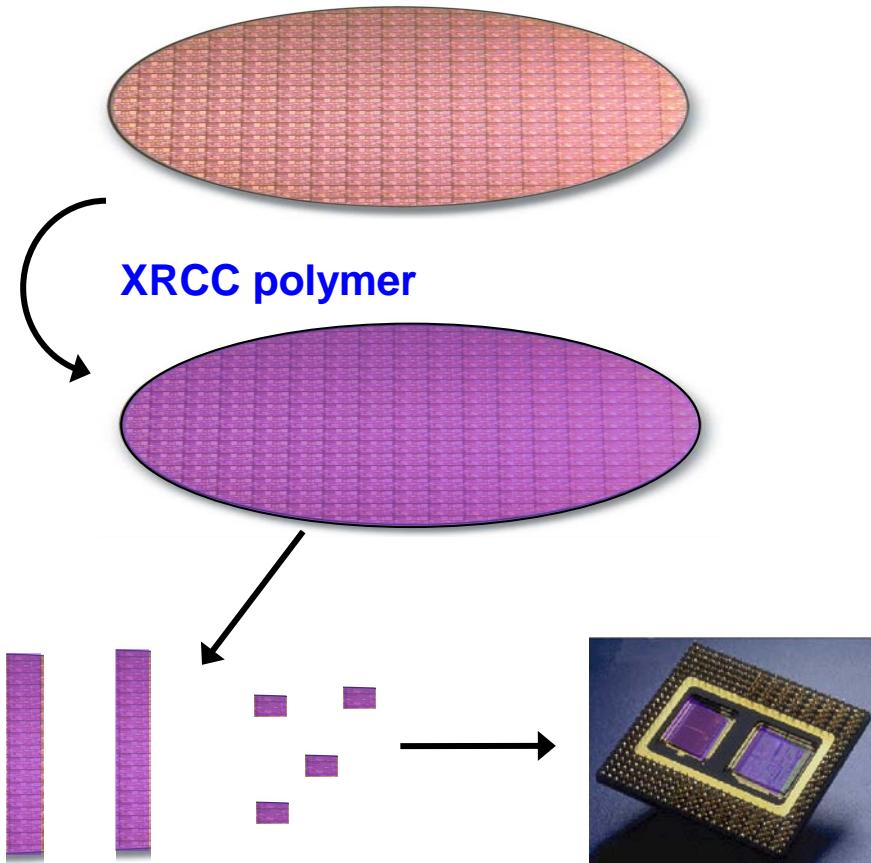


~100 X higher
“write” current
than drop cast,
ambient

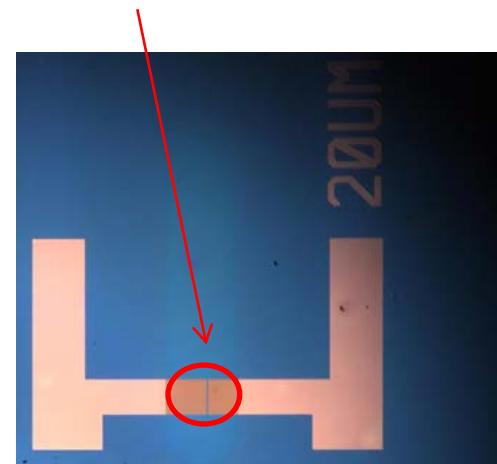


Note:
effective “write”
speed of ~2 msec
instead of ~500 msec

CMOS with support electronics



Thin film transistor substrate



Target: high “value added”
to CMOS by integrating
molecular nonvolatile
memory

Low density prototype tester:

