

# NMR III

## Practice of NMR

### Proton Equivalence

### Time Scale

**Ref**      9 : 3, 7, 9; 8<sup>th</sup> ed.  
              9 : 5, 8, 10; 9<sup>th</sup> ed.

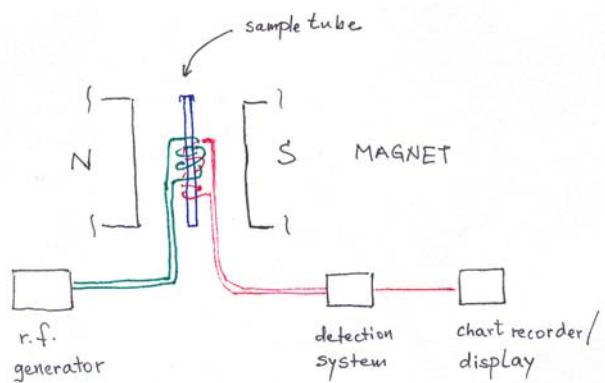
**Prob**      9 : 1 - 4; 8<sup>th</sup> ed.  
              9 : 3 - 6; 9<sup>th</sup> ed.

**HMWK #3**

**Adv Rdg** 9 : 8; 8<sup>th</sup> ed.  
              9 : 9; 9<sup>th</sup> ed.

## NMR Instrument

### Functional Diagram



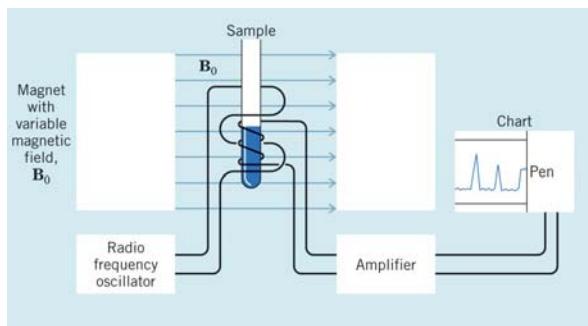
detection of resonance conditions

- “traditional”: fixed r.f.; vary  $B_0$
- “alternative”: fixed  $B_0$ , vary r.f.
- “modern”: “Fourier Transform Technology”, using fixed magnet  
*(broad shot of rf's, then see how it decays)*

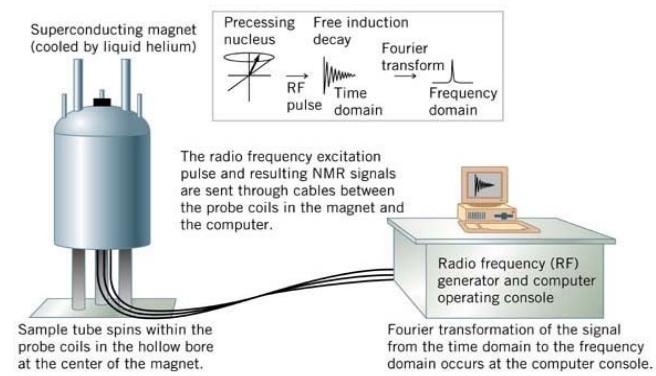
results always displayed as in “alternative”

for illustration, see .....

### Solomons 9.3, NMR Instrument, Traditional

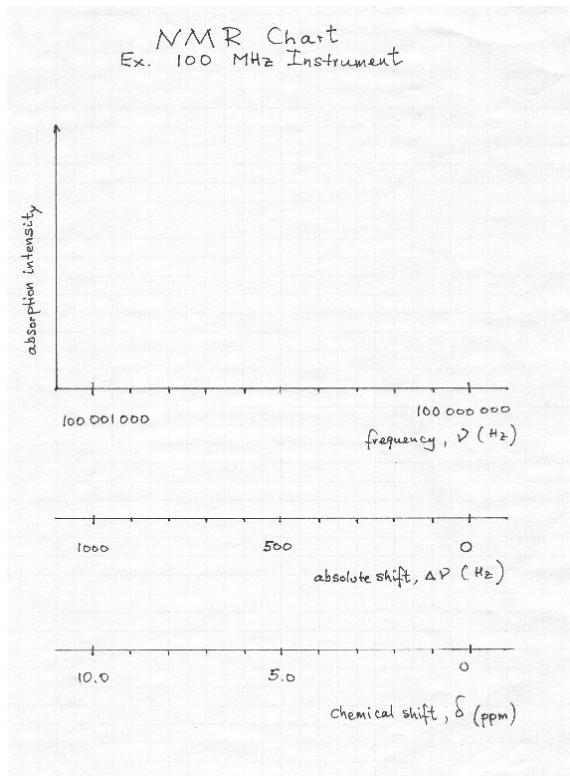


### Solomons 9.11, NMR Instrument, Modern



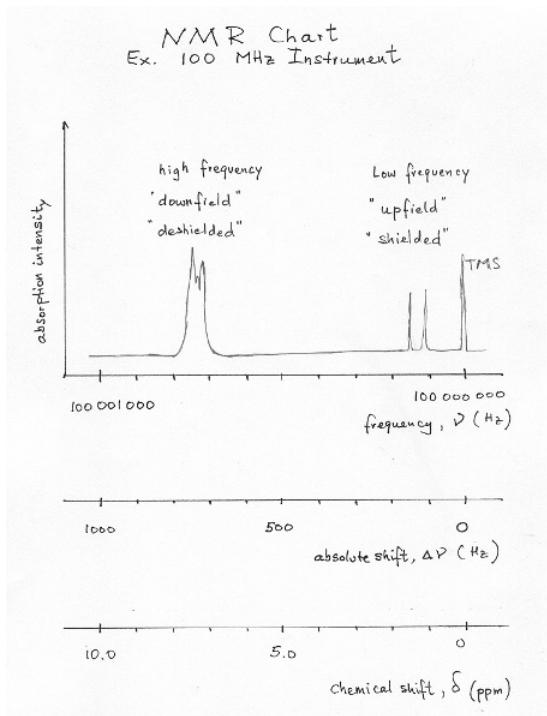
# NMR Chart

Ex. 100 MHz Instrument; Blank



# NMR Chart

Ex. 100 MHz Instrument; w/ Peaks



## Proton Equivalence

(Chem Equivalence only)

- identical (= "equiv") H's give 1 signal
- non-identical (= "non-equiv.") H's give diff. signals

### Implication

- equiv. H's have same chem. environment
- non equiv. ..... different .....

### Terminology:

Usually, several types of H's exist in a molecule,  
"each type" = "non-equivalent H"

## Practice, Non-Equivalent H's

cmpd	# of non-equivalent H's
$\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}-\text{H} \\   \\ \text{H} \end{array}$	1
$\begin{array}{cc} \text{H} & \text{H} \\   &   \\ \text{H}-\text{C} & -\text{C}-\text{H} \\   &   \\ \text{H} & \text{H} \end{array}$	1
$\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}=\text{C}-\text{H} \\   \\ \text{H} \end{array}$	1
$\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}=\text{C}-\text{H} \\   \\ \text{H} \end{array}$	3
$\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}(=\text{O})-\text{O}-\text{H} \\   \\ \text{H} \end{array}$	2
$\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}-\text{O}-\text{H} \\   \\ \text{H} \end{array}$	2

## Equivalence ...

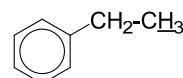
In difficult cases:

- replace H by “X”
- see if it gives different cmpd (*naming might help*)
- if get diff. enantiomers  
∴ doesn't count
- if get diff. diastereomers  
∴ counts as non-equivalent

## Terminology

if replacement gives      then H's are calledachiral identical cmpd      homotopic (*equiv.*)

Ex.:

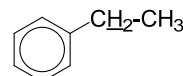


1 signal for 3 H's

enantiomers

enantiotopic (*equiv.*)

Ex.:

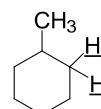


1 signal for 2 H's

diastereomers

diastereotopic (*non-equiv.*)

Ex.:

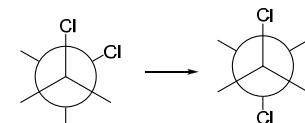
2 signals for 2 H's  
(*in theory*)

## Practice: Non-Equivalent Protons

cmpd	# of non-equivalent H's
	3
	3
	1
	4 (not counting $\text{CH}_3$ group)
	1 (enantiotopic)
	2 (diastereotopic)

## Time Scale

- NMR detection is slow
- cannot see individual conformations

e.g.,  $\text{Cl}-\text{CH}_2-\text{CH}_2-\text{Cl}$ conformational change not detectable,  
gives only 1 signal, reflecting avg. of conformations

Applies to:

- 1.) open chain systems

- 2.) mobile ring systems

e.g., cyclohexane



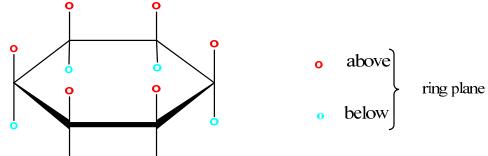
here, cannot differentiate between ax. &amp; eq.

Note:

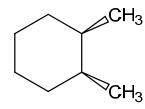
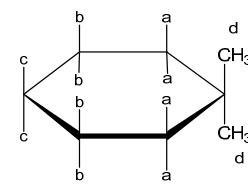
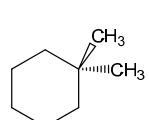
In rigid ring systems, such as trans decalins, steroids  
ax. & eq. can be differentiated

time scale

- NMR sees ring(s) (at r.t.) as planar structure, with H's above and below the plane
- better 3D visualisation for NMR as follows:



## Practice : Count Non-Equivalent Protons



6 signals

