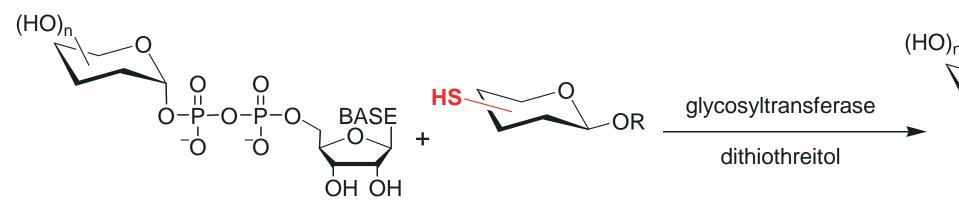


Jamie R. Rich, Adam Szpacenko, Monica M. Palcic and David R. Bundle Alberta Ingenuity Centre for Carbohydrate Science, Department of Chemistry, University of Alberta, Edmonton, Alberta, Canada, T6G 2G2

## **OVERVIEW**

The enzymatic formation of sulfur-linked oligosaccharides has been demonstrated by incubation of synthetic carbohydrate thiol acceptors with sugar nucleotide donors and an appropriate mammalian or bacterial glycosyltransferase. These represent the first glycosyltransferase catalyzed syntheses of thiooligosaccharides.



Scheme 1. Enzymatic formation of S-linked oligosaccharides.

### BACKGROUND

Oligosaccharide analogues in which the interglycosidic oxygen atom(s) have been replaced by sulfur are known as thiooligosaccharides. These molecules are metabolically stable analogues of their naturally occurring counterparts, since the rate of hydrolysis of the thioglycosidic bond by glycosylhydrolases is several orders of magnitude slower than that of the corresponding Oglycosides and the conformational space sampled by S-linked glycosides is similar to that of O-glycosides. In combination, these properties suggest that S-linked carbohydrates may be attractive antigens with extended in vivo activity,<sup>1</sup> and the ability of glycosyltransferases to create such linkages has significant potential.

Considerable attention has been focused on the synthesis of complex oligosaccharides containing S-linked residues and a variety of chemical approaches are available to construct this linkage.<sup>2</sup> Most thiooligosaccharide syntheses take advantage of the reduced basicity and enhanced nucleophilicity of sulfur compared to oxygen, employing a sugar thiol or thiolate anion in a reaction with another carbohydrate bearing an electrophilic leaving group. Recently active site mutants of glycosylhydrolases, termed "thioglycoligases" have been shown to transfer a glycosyl residue to sugar thiols.<sup>3</sup>

The development of a glycosyltransferase-based approach to the synthesis of thiooligosaccharides does not obviate the considerable effort required for the construction of thiosugar acceptors. However, this method demonstrates the high yields, stereo-, and regio-selectivities characteristic of glycosyltransferase catalyzed reactions, and renders a final deprotection step unnecessary

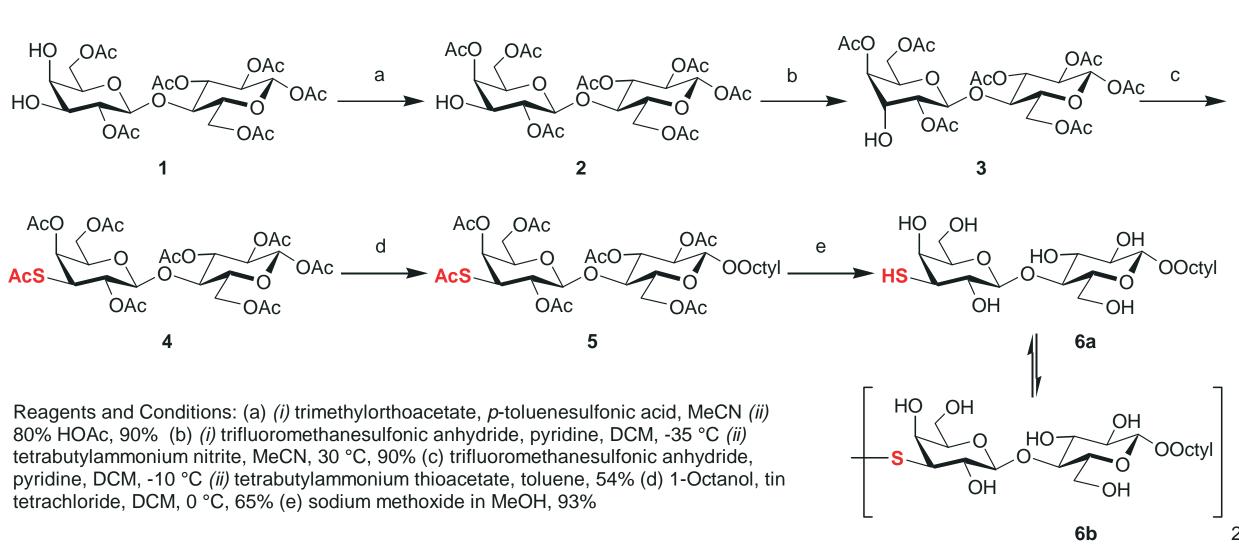
# SURVEYED GLYCOSYLTRANSFERASES

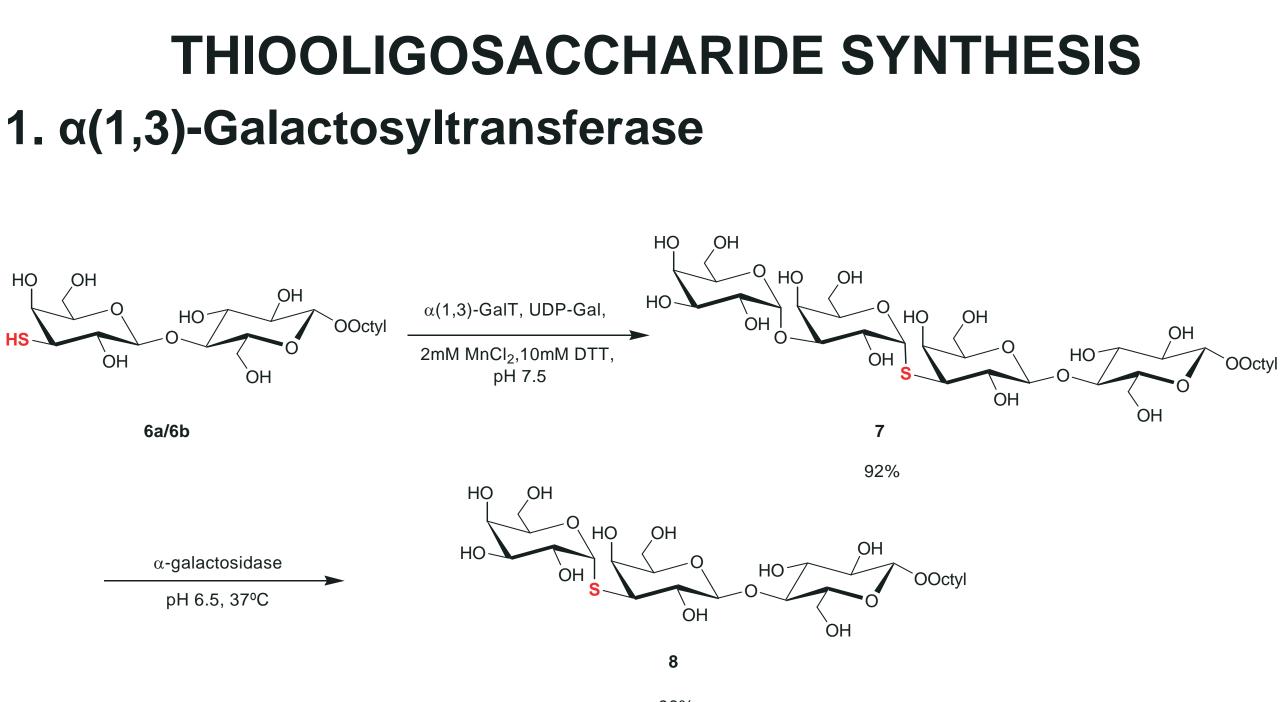
Enzyme	Source	Active?	Relative Rate
α (1,3) Galactosyltransferase <sup>4</sup>	Bovine	Y	0.2%
β (1,3) N-acetylglucosaminyltransferase <sup>5</sup>	N. Meningitidis*	Y	0.5%
α (2,3) Sialyltransferase	C. jejuni	Ν	-
ABO(H) blood group glycosyltransferase B (GTB)	Human	N **	-

**Table 1.** Glycosyltransferases were assayed for their ability to catalyze the synthesis of thiooligosaccharides via a standard radiochemical assay<sup>6</sup>. Rates are given as a % of the rate of transfer to the analogous unmodified acceptor.  $^*\beta(1,3)$ -GlcNAcT from another bacterial source was also shown to catalyze this reaction. \* \*The GTB-catalyzed reaction was conducted in the presence of an inhibitor which was not separated from the acceptor.

# **Glycosyltransferase Catalyzed Synthesis of Thiooligosaccharides**

# **REPRESENTATIVE SYNTHESIS OF A CARBOHYDRATE-THIOL ACCEPTOR**

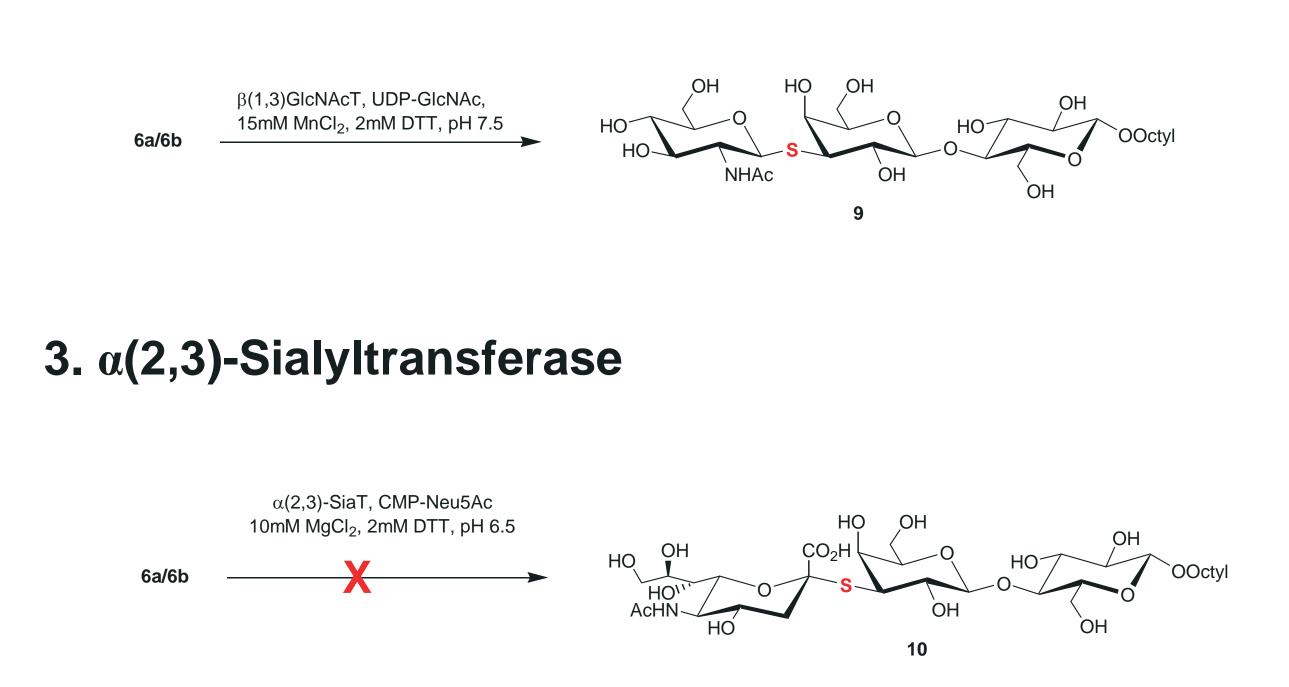




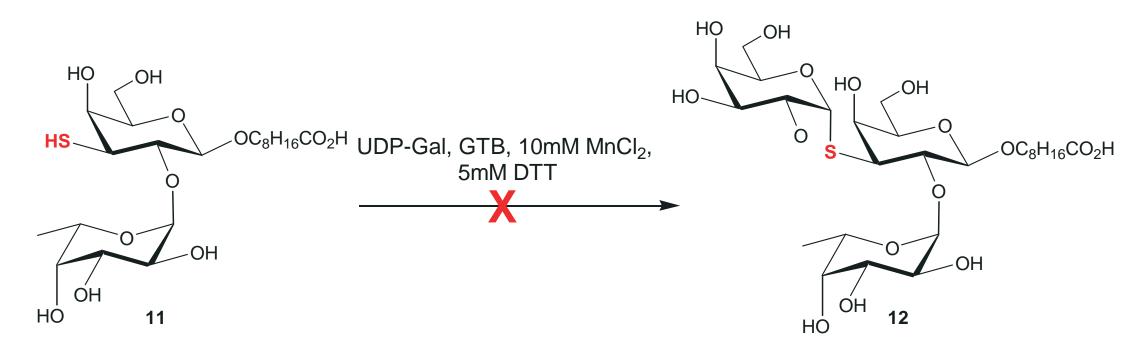
•Surprisingly, two  $\alpha$ -linked galactosyl residues were incorporated, yielding tetrasaccharide 7.

• Owing to the relative stability of the thioglycosidic linkage, the trisaccharide 8 was obtained as the sole product following treatment of 7 with an  $\alpha$ -galactosidase.

### **2.** $\beta$ (1,3)-*N*-acetylglucosaminyltransferase



#### 4. ABO(H) blood group glycosyltransferase B



Compound 11, a sulfur containing analogue of the H-disaccharide, was chemically synthesized in seven steps from known intermediates. A side product formed in the final deprotection step ( $\alpha$ -L-Fucp-(1-2)- $\beta$ -D-(3-deoxy)-Galp-(OR)) was not separated from **11**, and is a known inhibitor of GTB.

Both  $\alpha(1,3)$ -GalT and  $\beta(1,3)$ -GlcNAcT catalyze the glycosylation of thiol acceptor 6a, albeit at <1% of the rate of transfer to  $\beta$ -Octyl lactose. More detailed kinetic parameters have been determined for the reaction of acceptor **6a** catalyzed by  $\alpha(1,3)$ -GalT (Table 2). In light of the similarity of the K<sub>M</sub> values for **6a** and  $\beta$ -Octyl lactose and the substantially diminished rate of glycosyl transfer observed for the thio-analogue, it seems evident that the nucleophilicity of the thiol is greatly subdued in the active site.

Acceptor	V <sub>max</sub>	K <sub>M</sub>
β-Octyl lactose	1	5 mM
<b>6</b> a	0.002	15 mM
8	0.17	7 mM

**Table 2.** Kinetic parameters for the glycosylation of **6a** and **8** catalyzed by  $\alpha(1,3)$ -GalT

We have demonstrated a new method for the synthesis of sulfur-linked oligosaccharides using some bacterial and mammalian glycosyltransferases.

The observed rate of glycosyl transfer to thiols is diminished relative to transfer to natural acceptors.

As our understanding of glycosyltransferase activity increases, the development of mutants should permit the creation of redesigned enzymes with improved kinetics for the synthesis of S-linked oligosaccharides.

- 1. H.N. Yu. Doctoral Thesis, University of Alberta, 2002. Chem., 1997, 187, 85-116.
- Angew. Chem. 2003, 42, 352-354.
- Chemistry, 2002, pp. 127-134.

ACKNOWLEDGMENT: We wish to thank Irma van Die and Warren Wakarchuk for providing clones of  $\beta(1,3)$ GlcNAcT and  $\alpha(2,3)$ -SiaT. Funding was received from the Natural Sciences and Engineering Research Council of Canada (NSERC) (D.R.B., M.M.P.) and the Alberta Ingenuity Centre for Carbohydrate Science (D.R.B, M.M.P.). J.R. thanks NSERC and the Alberta Heritage Foundation for Medical Research for studentships.





#### **KINETICS**

#### CONCLUSIONS

#### REFERENCES

2. J. Defaye, J. Gelas, in Studies in Natural Products Chemistry. (Ed: Atta-ur-Rahman), Elsevier, Amsterdam, 1991, pp. 315-357; H. Driguez. *Chembiochem*, 2001, 2, 311-318; J.K. Fairweather, H. Driguez, in Carbohydrates in Chemistry and Biology, Vol 1. (Eds. B. Ernst, G.W. Hart, P. Sinaÿ), Wiley-VCH, 2000, pp. 531-564. ; H. Driguez, Top. Curr.

3. M. Jahn, J. Marles, R.A.J. Warren, S.G. Withers, Angew. Chem. 2003, 115, 366-368;

4. Y.R. Fang, K. Sujino, A. Lu, J. Gregson, R. Yeske, V.P. Kamath, R.M. Ratcliffe, M.J. Shur, W.W. Wakarchuk, M.M. Palcic, in *Carbohydrate Bioengineering, Interdisciplinary* Approaches. (Eds.: T.T. Teeri, B. Svensson, H.J. Gilbert, T.Feizi), Royal Society of

5. O. Blixt, I. van Die, T. Norberg, D.H. van den Eijnden, *Glycobiology* 1999, 9, 1061-1071. 6. M.M. Palcic, L.D. Heerze, M. Pierce, O. Hindsgaul, *Glycoconj. J.* 1988, *5*, 49-63.